

2-2 Reteach to Build Understanding

Standard Form of a Quadratic Function

1. What is the graph of $f(x) = 2x^2 - 8x + 5$? Fill in the blanks.

$a = 2$; $b = \underline{-8}$; $c = \underline{5}$

Find the equation of the axis of symmetry. $x = -\frac{b}{2a} = -\frac{(-8)}{2(2)} = \frac{8}{4} = 2$

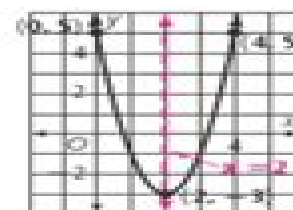
Find the x -coordinate of the vertex: $-\frac{b}{2a} = \underline{2}$

Find the y -value when $x = 2$. $f(x) = 2(2)^2 - 8(2) + 5$
 $= \underline{8 - 16} + 5 = -3$

y -coordinate of vertex: -3 The vertex is $(2, \underline{-3})$.

y -intercept: $(0, \underline{5})$ The y -intercept is at $(0, c) = (0, \underline{5})$.

Because a is positive, the graph opens upward, and the vertex is at the bottom of the graph. Plot the vertex and draw the axis of symmetry. Plot $(0, 5)$ and its corresponding point on the other side of the axis of symmetry.



2. Abby found the axis of symmetry, x -coordinate, the vertex, and y -intercept of the equation $f(x) = -x^2 - 8x - 15$. Find and explain Abby's errors.

axis of symmetry $x = -4$

x -coordinate of vertex $x = -4$

vertex: $(4, -31)$

y -intercept $(0, -15)$

The vertex should be $(1, -4)$.

Abby left off the negative sign when making the calculations for the vertex.

3. Find the key features of the following equation, which is in standard form:

$f(x) = 2x^2 + 8x - 3$

a. Find the equation of the axis of symmetry. $x = -\frac{b}{2a} = \frac{-8}{2(2)} = \frac{-8}{4} = \underline{-2}$

b. Find the x -coordinate of the vertex: $-\frac{b}{2a} = \underline{-2}$

c. Find the y -value when $x = -2$. $f(x) = 2x^2 + 8x - 3$
 $f(2) = 2(-2)^2 + 8(-2) - 3$
 $= \underline{8 - 16 - 3} = \underline{-11}$

d. y -coordinate of vertex: -11 The vertex is $(\underline{-2}, \underline{-11})$.

e. The y -intercept is at $(0, c) = (\underline{0}, \underline{-3})$.

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and

protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus