

KEY

Chemistry Worksheet: Matter #1

1. A mixture (~~is~~ is not) a chemical combining of substances.
2. In a compound the (atoms ~~molecules~~) are (chemically ~~physically~~) combined so that the elements that make up the compound (retain ~~lose~~) their identities and (~~do~~ do not) take on a new set of properties.
3. The smallest identifiable unit of a compound is a(n) molecule, which is made up of atoms _{elements} which are chemically bonded.
4. True or False: A mixture is always made up of a combination of elements. *or compounds*
5. In a mixture, the substances (~~lose~~ retain) their identities.
6. In a mixture the substances involved (can ~~cannot~~) be separated by a simple physical process.
In a compound the elements involved (can ~~cannot~~) be separated by a simple physical process because the elements are (physically combined ~~chemically bonded~~).
7. True or False: An element can be broken down into a simpler substance.
8. The smallest identifiable unit of an element is a(n) atom.
9. From the following list of substances, circle the ones that are elements:

<u>silver</u>	carbon dioxide	wood alcohol	<u>chromium</u>
<u>water</u>	<u>hydrogen</u>	<u>carbon</u>	<u>nitrogen</u>
<u>oxygen</u>	<u>gold</u>	sugar	salt
air	<u>sulfur</u>	<u>magnesium</u>	<u>nickel</u>
10. Explain how to separate the sugar and water in a solution of sugar and water.
Boil the water to leave the sugar behind.
11. How would you separate a mixture of alcohol and water?
Freeze the mixture, water will freeze and alcohol will remain a liquid.
12. How would you separate sand and water?

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Daniel C. Pantaleo



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