

## Chapter 9 Carbon Chemistry

**Section 9.1 Carbon Compounds****(pages 252–259)**

This section describes different forms of carbon that exist in nature. It also discusses saturated and unsaturated hydrocarbons. It explains the formation of fossil fuels and describes the products of their combustion.

**Reading Strategy (page 252)**

**Preliminary** Before you read, use the models in Figure 2 to describe the arrangement of carbon atoms in each form of carbon. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

| Forms of Carbon      |  |
|----------------------|--|
| Diamond              |  |
| Graphite             |  |
| Buckminsterfullerene |  |

- The two elements that all organic compounds contain are \_\_\_\_\_.
- Circle the letter of the approximate percentage of all known compounds that are organic compounds.
  - 10 percent
  - 30 percent
  - 60 percent
  - 90 percent

**Forms of Carbon (page 253)**

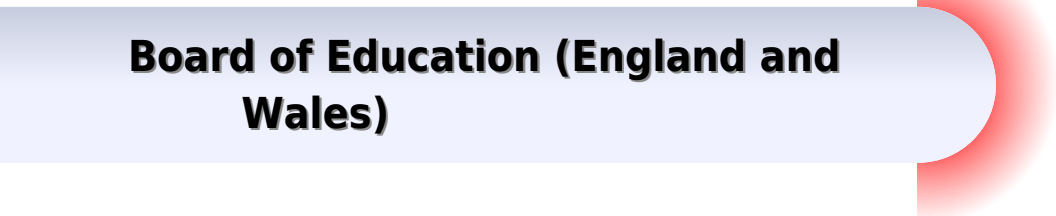
- Circle the letter of each form of carbon.
  - soot
  - diamond
  - fullerenes
  - graphite
- Describe a network solid. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letter of each property of graphite.
  - soft
  - rigid
  - compact
  - slippery

**Saturated Hydrocarbons (pages 254–255)**

- Is the following sentence true or false? A hydrocarbon is an organic compound that contains carbon, hydrogen, and oxygen. \_\_\_\_\_
- Is the following sentence true or false? A saturated hydrocarbon contains only single bonds. \_\_\_\_\_

# Section 1 Carbon Compounds Answers

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Wales)**



## Section 1 Carbon Compounds Answers:

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