

practice test ch 7 & 8

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following best describes the rates of chemical reaction?
- most chemical reactions occur very slowly
 - most chemical reactions occur at moderate rates
 - most chemical reactions occur very rapidly
 - chemical reactions have a wide range of rates, from extremely fast to extremely slow
- _____ 2. Which of the following is true of the rates of most chemical reactions?
- the initial rate is cannot be measured
 - the initial rate is faster than the rate later in time
 - the initial rate is the same as the rate later in time
 - the initial rate is slower than the rate later in time
- _____ 3. Which of the following is true of effective collisions?
- the number of effective collisions decreases as the temperature is increased
 - the number of effective collisions determines the reaction rate
 - both a. and b.
 - Neither a. nor b.
- _____ 4. Which of the following is true of effective collisions?
- the number of effective collisions determines the reaction rate
 - the number of effective collisions increases as the temperature is increased
 - both a. and b.
 - Neither a. nor b.
- _____ 5. Which of the following is true?
- decreasing the activation increases the reaction rate
 - the activation energy for an exothermic reaction is negative
 - both a. and b.
 - Neither a. nor b.
- _____ 6. In an energy diagram for a chemical reaction what species appears at the highest energy?
- the catalyst
 - the products
 - the reactants
 - the transition state
- _____ 7. Given that the reaction $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$ is exothermic, which of the following is true of the reaction $2 \text{H}_2\text{O}(\text{g}) \rightarrow 2 \text{H}_2(\text{g}) + \text{O}_2(\text{g})$?
- its activation energy is lower than that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is higher than that of c. its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - there is no relationship between its activation energy and that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
- _____ 8. In an energy diagram for an endothermic chemical reaction which of the following is true?
- the energy of the products is lower than that of the reactants
 - the energy of the transition state is higher than that of the reactants
 - both a. and b.
 - Neither a. nor b.
- _____ 9. Consider a reaction such as $\text{A}_2(\text{g}) + \text{B}_2(\text{g}) \rightarrow 2\text{AB}(\text{g})$. What can we say about the mechanism of the reaction?
- if the reaction is fast it occurs in a single step
 - if the reaction is fast it occurs in multiple steps
 - if the reaction is slow it occurs in multiple steps

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Jicheng Xie



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Holt Modern Chemistry Chapter 7 Test Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Holt Modern Chemistry Chapter 7 Test Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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