Name: ______ Period: _____ Date: _____

Egyptian Civilization Matching

Match the definition from the column on the right with the corresponding term in the column on the left. Place the letter of the definition on the blank next to the correct term.

- L 1. Aten
 P 2. Black Land
- T 3. Canal
- O 4. Chariots
- N 5. Civil
- M 6. Civilization
- E 7. Dynasty
- K 8. Hyksos
- G 9. Khufu
- A 10. Kingdoms
- 5 11 Linen
- ___B___12. Lower
- R 13. Mountains
- ____C___14. Nille
- _______15. Nubia
- __F__16. Pyramid
- ____ 17. Red Land
- ____ D___ 18. Red Sea
- H 19. Scribes

- A. The history of ancient Egypt is divided into three ______.
- B. The part of ancient Egypt located in the northern delta where the river spreads out and empties into the Mediterranean Sea was called ______ Egypt.
- C. The large river along which the Egyptian Civilization developed
- D. The sea to the east of the Nile River
- E. A series of rulers from the same family or line
- F. An Egyptian tomb with a rectangular base and four triangle faces that meet at the top
- G. The pharaoh buried in the Great Pyramid of Giza
- H. People who kept records and wrote messages
- King Mentuhotep made this city the capital during the Middle Kingdom
- The country south of Egypt from which Egyptians brought gold, ivory, and slaves
- K. Egypt was conquered by these people from Canaan. Their victory over the Egyptians marked the end of the Middle Kingdom
- The name of the sun god Pharaoh Amenhotep IV forced his people to worship
- M. The culture developed by a particular region or nation
- N. A war between two or more groups in the same country
- O. Horse-drawn vehicles used in war
- P. The Egyptians called their country the _______because of the fertile soil
- Q. The Egyptians called the desert surrounding their country the
- R. The Nile River is fed by rain and melting snow from the _____ of northeastern Africa
- S. Rich Egyptians were wrapped in this material when they died
- The Egyptians built one of these between the Nile and the Red Sea in order to improve trade

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution ofcopper ii sulfate lab report 4 stoichiometry of iron copper ii sulfate - Feb 09 2023

web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web lab 4 stoichiometry the reaction of iron with copper ii sulfate revised 8 19 2009 1 introduction in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with

a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated cuso4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g