

Double-Replacement Precipitation Lab

Introduction: Double-Replacement reactions occur when two ionic solutions mix to form an insoluble solid. If all cations and anions are soluble then no reaction will occur. During the lab procedure, several ionic solutions will be mixed to determine if a precipitate reaction will occur. Those reactions that do not form a precipitate will be marked for No Reaction (NR). Students will practice writing and balancing double-replacement reactions.

Pre-Lab questions:

1. How would you describe a double replacement reaction to a friend who is struggling with reaction types?
2. What is a precipitate?

Materials:

- 0.2M sodium bromide, NaBr
- 0.2M sodium carbonate, Na_2CO_3
- 0.2M sodium phosphate, Na_3PO_4
- 0.2M sodium sulfate, Na_2SO_4
- 0.2M sodium hydroxide, NaOH
- Transparency paper
- 0.2M lithium chloride, LiCl
- 0.2M copper(II) chloride, CuCl_2
- 0.2M zinc chloride, ZnCl_2
- 0.2M strontium chloride, SrCl_2
- 0.2M barium chloride BaCl_2
- Background paper Template

Safety: You MUST wear goggles during this lab activity

Procedure:

1. Place the overhead sheet over the background paper.
2. Using the Data Table as a guide, mix every possible pair of solutions in a separate square of the background paper. **Use 1 or 2 drops of each solution.** Do NOT contaminate the individual pipets with different solutions.
3. Mix the solutions with a rinsed toothpick if you are uncertain if a precipitate formed. Rinse the toothpick after EACH mixing – be careful of which end of the rod you use so that you do not touch the chemicals or contaminate your samples!
4. Observe each mixture carefully for signs of a precipitate. (Some precipitates are light in color and hard to see so you may want to slide the transparency over the black lab top on occasion.) Record your observations (presence or absence of a precipitate, and other characteristics such as color, texture, etc.) in the Data Table. If nothing happens, write NR for no reaction.
5. Rinse and wipe the transparency sheet into the sink. Dry the sheet.

Observations: Write the observations you made from combining these materials in the spaces in the table below. Note any color changes or precipitate (cloudiness/solid). If no change is observed write NR for no reaction.

	NaBr	Na_2CO_3	Na_3PO_4	Na_2SO_4	NaOH
LiCl					
CuCl_2					
ZnCl_2					
SrCl_2					
BaCl_2					

Double Replacement Reaction Lab Conclusion Answers

Kristin Tuttle Bump



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Safety-Scale Lab Exp Biochem 2e Spencer L. Seager, Michael R. Slabaugh, 1994-05 Steps to Doing Science Kristin Tuttle Bump, 2008 **Inquiry-based Experiments in Chemistry** Valerie Ludwig Lechtanski, 2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student Novice Teacher Action Anne Liu Kern, 2007 Everything You Need to Ace Chemistry in One Big Fat Notebook Workman Publishing, Jennifer Swanson, 2020-09-01 Chemistry No problem This Big Fat Notebook covers everything you need to know during a year of high school chemistry class breaking down one big bad subject into accessible units Learn to study better and get better grades using mnemonic devices definitions diagrams educational doodles and quizzes to recap it all Including Atoms elements compounds and mixtures The periodic table Quantum theory Bonding The mole Chemical reactions and calculations Gas laws Solubility pH scale Titrations Le Chatelier's principle and much more **Prentice Hall Chemistry**, 2000 **Authentic Practice Chemistry** David G. White, 2000 Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 **Harcourt Science: Life science, units A and B**, 2002

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boxrob 3 truck loader 4 truck loader 3 traktör oyunları bedava online traktör oyunlarımız tarlaları sürecekle çöpleri ezecek ve tarla aletlerini canavar kamyonlar gibi kullanacak sürücülerini arıyor Ücretsiz traktör oyunlarımızdan birini seç ve eğlen