

HEAT TRANSFER AND FLUID FLOW (THERMODYNAMICS) EXAM QUESTIONS WITH CORRECT ANSWERS

Potential Energy - ANSWER-is the stored energy of position, and exists due to vertical distance (z) over which gravitational forces can be exerted on a system's mass.

$PE = mgz$ where: m = mass, g = gravitational constant, and z = vertical height of mass.

Kinetic Energy - ANSWER-is the mass's energy of motion.

$KE = mv^2 / 2$ where: m = mass, v = velocity of fluid

Flow Energy - ANSWER-is the work done by the working fluid on its surroundings as it moves from one point to another in a system.

$FE = PV$ where: P = pressure, V = volume

Internal Energy - ANSWER-(U) is the sum of energy in a substance. A substance possesses several microscopic forms of energy due to rotation, vibration, translation, and interactions among molecules of the substance.

Enthalpy (H) - ANSWER-Enthalpy is the amount of heat content used or released in a system at constant pressure.

$h = (PV + U) / m$ (BTU/lbm)

Second Law of Thermodynamics - ANSWER-No heat engine, actual or ideal, when operating in a cycle can convert all the heat supplied to it into mechanical work. (i.e., you cannot have a 100% efficient cycle).

Entropy - ANSWER-When considering how much work can be produced with an amount of heat, we must know the temperature at which heat is transferred in addition to the amount of heat transfer. Entropy is a property used to describe both the amount of heat transferred and the temperature at which it is transferred. Change in entropy is normally defined instead of entropy itself.

Explain the General Energy Equation - ANSWER- $(KE + PE + FE + U + Q)_{in} = (KE + PE + FE + U + W)_{out} + (KE + PE + FE + U)_{stored}$

Steam Tables Consist of Three Sets of Tables - ANSWER-Table 1 - Saturated Steam Table By Temperature
Table 2 - Saturated Steam Table By Pressure

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VM Jensen



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