# PRINICIPLE OF PHYSICAL CHEMISTRY

PURI SHARMA PATHANIA PATHANIA

# **Physical Chemistry By Puri Sharma Pathania**

**Dr. J Harihara Reddy** 

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molar quantities free energy volume heat concept Gibb's Duhem equation Chapter 3 Chemical Dynamics I Effect of temperature on reaction rates Rate law for opposing reactions of 1st order and 1Ind order Rate law for consecutive Collision theory of reaction rates and its limitations Steric factor Activated complex theory Ionic reactions single and double sphere models Influence of solvent and ionic strength The comparison of collision and activated complex theory Chapter 4 Electrochemistry I Ion Ion Interactions The Debye Huckel theory of ion ion interactions Potential and excess charge density as a function of distance from the central ion Debye Huckel reciprocal length Ionic cloud and its contribution to the total potential Debye Huckel limiting law of activity coefficients and its limitations Ion size effect on potential Ion size parameter and the theoretical mean activity coefficient in the case of ionic clouds with finite sized ions Debye Huckel Onsager treatment for aqueous solutions and its limitations Debye Huckel Onsager theory for non aqueous solutions The solvent effect on the mobality at infinite dilution Equivalent conductivity vs concentration c 1 2 as a function of the solvent Effect of ion association upon conductivity Debye Huckel Bjerrum equation Chapter 5 Quantum Mechanics II Schrodinger wave equation for a particle in a three dimensional box The concept of degeneracy among energy levels for a particle in three dimensional box Schrodinger wave equation for a linear harmonic oscillator Zero point energy of a particle possessing harmonic motion and its consequence Schrodinger wave equation for three dimensional Rigid rotator Energy of rigid rotator Space quantization Schrodinger wave equation for hydrogen atom separation of variable in polar spherical coordinates and its solution Principle azimuthal and magnetic quantum numbers and the magnitude of their values Probability distribution function Radial distribution function Shape of atomic orbitals s p Law of mass action and its thermodynamic derivation Third law of thermodynamics Nernest heat theorem determination of absolute entropy unattainability of absolute zero and its limitation Phase diagram for two completely miscible components systems Eutectic systems Calculation of eutectic point Systems forming solid compounds Ax By with congruent and incongruent melting points Phase diagram and thermodynamic treatment of solid solutions Chapter 7 Chemical Dynamics II Chain reactions hydrogen bromine reaction pyrolysis of acetaldehyde decomposition of ethane Photochemical reactions hydrogen bromine General treatment of chain reactions ortho para hydrogen conversion and hydrogen bromine reactions Apparent activation energy of chain reactions Chain length Rice Herzfeld mechanism of organic molecules decomposition acetaldehyde Branching chain reactions and explosions H2 O2 reaction Kinetics of one intermediate enzymatic reaction Michaelis Menton treatment Evaluation of Michaelis s constant for enzyme substrate binding by Lineweaver Burk plot and Eadie Hofstae methods Competitive and non competitive inhibition Chapter 8 Electrochemistry II Ion Transport in Solutions Ionic movement under the influence of an electric field Mobility of ions Ionic drift velocity and its relation with current density Einstein relation between the absolute mobility and diffusion coefficient The Stokes Einstein relation The Nernst Einstein equation Walden's rule The Rate process approach to ionic migration The Rate process equation for equivalent conductivity Total driving force for ionic transport Nernst Planck Flux

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Advanced Physical Chemistry DN Bajpai,2001-01-21 A Textbook for B Sc Part III and Hons and Postgraduate Courses of Indian Universities In this edition I have made major changes in the light of modern concepts introduced in syllabi at the under graduate and postgraduate level as well With matter has also been updated The subject matter has been arranged systematically in a lucid style and simple language New Problems and exercises have also been introduced to acquaint the students with trend of questions they except in the examinations 
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Physical Chemistry M. V. Sangaranarayanan,V. Mahadevan,2012-04-16 Textbook of Physical Chemistry together with the companion books on organic chemistry and inorganic chemistry meets the complete requirements of undergraduate students of chemistry across India In a book comprising all the classical topics which span physical chemistry including chemical kinetics electrochemistry and

thermodynamics among others uniformity in the depth of coverage of each topic is not easy to attain in view of the disjointed pace of growth of each discipline Nevertheless care has been taken to ensure that the material in this book will sustain the interest of students and motivate them to learn physical chemistry In order to aid students every chapter contains the Objectives at the beginning and Key Points at the end Various aspects of physical chemistry are dealt with in a lucid manner and interesting related matter is highlighted in boxes The derivations are given in a comprehensible manner Since physical chemistry involves numericals several worked examples complement the text The exercises at the end of each chapter in particular will be extremely valuable to sharpen the problem solving skills and direct the student towards appreciating the nuances of physical chemistry

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#### Physical Chemistry By Puri Sharma Pathania Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Physical Chemistry By Puri Sharma Pathania**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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