PART I- TRANSPORT PROCESSES: MOMENTUM, HEAT, AND MASS (0.2 - 1)°C= 80.05= 80°C Eq. (1.3-4) K = 353.2 = °C+ 273.15 176°F of a °C=80.0=(1/1.8)(°F-32) Eq. (1.3-2) 636°R * R = Eq. (1.3-3) OR = *F+ 460=176+460 (1, 2-2)Eq.(1.3-2) °C = (1/1.8)(°F-32)=(1/1.8)(155-32) = 68.33°C Eq. (1.3-4) K = °C + 273,15 = 68.33+273,15 = 341.5 K 615 °R Eq. (1.3-3) °R = °F + 460 = 155 + 460 = (1.3-1) MW(02)= 32.00, MW(N2)= 28.02 MW (air) = 0.21(32.00) + 0.79(28.02) = 28.9 de /deg mod (1.3-2) 200 + 02 -> 2002 Moles CO = 56.0 kg / kg mol = 2.00 kg mol MW 28.01 32.00 44.01 moles 02 = 1.00, 1.00(32.00) = 32.0 kg 02 2.00(44.0) = 88.0 kg CO2 (1.3-3) Das MW a mol mol her N2 0.1648 molpe 20 28.02 0.7138

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