

## Balancing Equations & Types of Reactions WS 1

### Practice Problems

In your notebook, solve the following problems.

NOTE: A skeleton equation is an unbalanced equation.

#### SECTION 10.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.



2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.



3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.



4. Write a balanced equation for the production of HCl gas from its elements.



5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



Hydrochloric acid and calcium carbonate react to produce carbon dioxide and calcium chloride and water.

6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



2 solid silver reacts w/ 1 solid sulfur to create solid silver sulfide.

# Describing Chemical Reactions Section Review Answer Key

**Holt Rinehart & Winston**



## **Describing Chemical Reactions Section Review Answer Key:**

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