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Moseley's Periodic Table

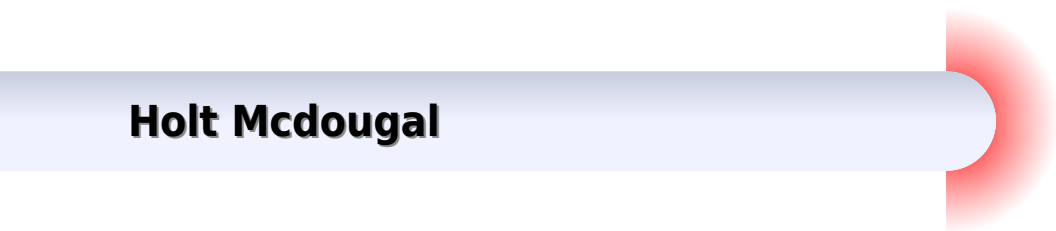
He studied (1909) the frequency of the X-ray produced by the bombardment of a strong beam of electrons on 'metal target.

He found that the square root of the frequency of X-rays ($\sqrt{\nu}$) is directly proportional to number of effective nuclear charge (z) of metal i.e. to atomic number and not to atomic mass of the atom of that metal.(as nuclear charge of metal atom is equal to atomic number)

i.e. ($\sqrt{\nu}$) = $a(z - b)$

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