

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

C.2.2 a 1. cationC.2.2 g 2. electronegativityC.2.5 j 3. inner transition metalC.2.6 i 4. alkali earth metalC.2.6 h 5. noble gasC.2.5 b 6. representative elementC.2.2 d 7. metalC.2.6 f 8. halogenC.2.5 e 9. transition metalC.2.2 c 10. ionization energy

a. a positively charged ion

b. an element whose highest occupied *s* or *p* sublevels are partially filled

c. the energy required to remove an electron from an atom

d. good conductor of heat and electric current

e. any element whose highest occupied *s* sublevel and a nearby *d* sublevel contain electrons

f. any nonmetal in group 7A

g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron)

h. an element in which the highest occupied *s* and *p* sublevels are completely filled

i. an element in group 2A

j. an element whose highest *s* sublevel and a nearby *f* sublevel contain electrons

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

C.2.6 b 11. The alkali metals do not include

a. Li

b. Ca

c. Na

d. Rb

C.2.6 c 12. The element iodine, I, is a

a. period 5 alkali metal

b. period 4 halogen

c. period 5 halogen

d. period 5 transition metal

Modern Chemistry Chapter 6 Test Answer Key

Charles Meymott Tidy



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Lab Experiments Modern Chemistry Harold Clark Metcalfe, 1982 **Laboratory Experiments to Accompany "Modern Chemistry,"** Fredus Nelson Peters, 1902 **Teacher's Manual and Resource Guide for Exploring the Sciences** Herbert Drapkin, 1964 **Modern inorganic chemistry** Joseph William Mellor, 1918 *Elementary Modern Chemistry* Wilhelm Ostwald, Harry Wheeler Morse, 1909 *Modern Environmental Analysis Techniques for Pollutants* Chaudhery Mustansar Hussain, Rustem Kecili, 2019-08-20

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