

2.5

ELEMENTS AND COMPOUNDS

Section Review

Objectives

1. Explain the difference between elements and a compound
2. Distinguish between a mixture and a solution
3. Identify the chemical symbols for elements and write a chemical formula for a compound

Vocabulary

- | | |
|------------|----------------|
| 1. element | 2. compound |
| 3. mixture | 4. homogeneous |

Part A Completion

Fill in the blank with the word or phrase that will complete the statement correctly. Write your answer in the space provided. Write the chemical symbol for the element in parentheses.

- | | |
|---|-------------------|
| 1. An element is made of _____ atoms. | 6. Element |
| 2. Compounds are made of _____ atoms. | 7. compound |
| 3. An element is made of _____ atoms. | 8. mixture |
| 4. Compounds are made of _____ atoms. | 9. homogeneous |
| 5. Compounds are _____. | 10. element |
| 6. The composition of a compound is _____. | 11. mixture |
| 7. The composition of a compound is _____. | 12. heterogeneous |
| 8. The composition of a compound is _____. | 13. element |
| 9. The composition of a compound is _____. | 14. heterogeneous |
| 10. The composition of a compound is _____. | 15. element |
| 11. The composition of a compound is _____. | 16. heterogeneous |
| 12. The composition of a compound is _____. | 17. element |
| 13. The composition of a compound is _____. | 18. heterogeneous |
| 14. The composition of a compound is _____. | 19. element |
| 15. The composition of a compound is _____. | 20. heterogeneous |

3 Elements And Compounds Section Review Answer Key

Kaplan Test Prep

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3 Elements And Compounds Section Review Answer Key:

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