# 4.1 EXTERNAL ASSESSMENT 2020: CHEMISTRY PAPER 1

#### Time allowed

- Persoal time 10 minutes
- Working time 90 minutes

#### General instructions

- Answer all questions in this question and response book.
- QCAA-approved calculator permitted.
- QCAA formula and data book psyvided.
- · Planning paper will not be marked.

#### Section 1 (20 marks)

20 multiple choice questions

#### Section 2 (40 marks)

7 short response questions

6 Fair of Queenland (QCAK) External assessment 2005, Chemistry Paper 1

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# Section 1 (20 marks)

# MULTIPLE CHOICE QUESTIONS



#### QUESTION I

A purify filled water bottle is socied and left on a bench in a room with a constant temperature. After several minures, it is need that the water level in the bottle remains constant. In the water bottle, the rate of emporation is

- (A) line than the rate of condensation.
- (III) around than the upp of condensation.
- (C) equal to the rate of condensation and equal to zone.
- (D) egod to the rate of combeniation but not equal to zero.

#### QUESTION 2

$$Mu(i) + 2Au'(ini) \rightarrow 2Au(i) + Mu''(ini)$$

Determine which of the following statements is true for the chemical reaction.

- (A) Silver ions are reduced and act as a reducing agent.
- (B) Silver loss are enidled and set as a reducing spent.
- (C) Magnesium atoms are reduced and act as a reducing agent.
- (D) Magnesium atoms are enableed and act as a reducing agent.

#### QUESTION 3



This general chemical equation represents the following type of reaction.

- CAC addition.
- (The Brodenhoise
- (C) experification
- The condensation





# **123 Section Assessment Answers Chemistry**

**Rosina Ehmann** 

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Basic Concepts of Chemistry, Study Guide Leo J. Malone,1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples *Complete Phlebotomy Exam Review E-Book* Pamela Primrose,2010-03-05 Complete Phlebotomy Exam Review contains 1000 questions and a wealth of content review to prepare you for the phlebotomy certification exams a mock certification exam at the end of the book tests your knowledge of necessary information This title includes additional digital media when purchased in print format For this digital book edition media content is not included Practice questions with rationales explain the correct answer and break the subject matter into manageable areas Mock certification exam gives you a chance to see how well you know the material

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