

# 14-1 Review and Reinforcement (continued)

Complete each of the sentences below.

15. The stronger the intermolecular forces in a liquid, the \_\_\_\_\_ its boiling point.
16. The state of a substance at room temperature depends on \_\_\_\_\_.
17. Water molecules in an ice cube are held together by \_\_\_\_\_ forces or, more specifically, \_\_\_\_\_ bonds.

On the line at the left, write the letter of the term that matches each description below. Each choice can be used once, more than once, or not at all.

a. dispersion

b. dipole-dipole

c. hydrogen bonding

- \_\_\_\_\_ 18. the force in between  $\text{SO}_2$  molecules
- \_\_\_\_\_ 19. the force that accounts for HF being a liquid while  $\text{H}_2$  and  $\text{F}_2$  are gases
- \_\_\_\_\_ 20. the force that depends on temporary, induced dipoles
- \_\_\_\_\_ 21. the only intermolecular force that exists in noble gases
- \_\_\_\_\_ 22. the attractive force between neighboring dipoles
- \_\_\_\_\_ 23. the force that arises because of the large difference in electronegativities in the N-H bonds

Answer the following question as directed.

24. Which of the following statements could be used to explain why water has an unusually high boiling point? Make a checkmark on the line for all that apply.
- |       |                                                                                  |
|-------|----------------------------------------------------------------------------------|
| _____ | a. Water molecules have a strong intermolecular force.                           |
| _____ | b. There are no nitrogen atoms in water.                                         |
| _____ | c. Oxygen is more electronegative than hydrogen.                                 |
| _____ | d. Water is an angular molecule.                                                 |
| _____ | e. Oxygen atoms are smaller than hydrogen atoms.                                 |
| _____ | f. Water does not contain metallic bonds.                                        |
| _____ | g. There is a large difference in electronegativity between oxygen and hydrogen. |

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## **Table of Contents 14 3 Review And Reinforcement Chemistry Answers**

1. Understanding the eBook 14 3 Review And Reinforcement Chemistry Answers
  - The Rise of Digital Reading 14 3 Review And Reinforcement Chemistry Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying 14 3 Review And Reinforcement Chemistry Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 14 3 Review And Reinforcement Chemistry Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from 14 3 Review And Reinforcement Chemistry Answers
  - Personalized Recommendations
  - 14 3 Review And Reinforcement Chemistry Answers User Reviews and Ratings
  - 14 3 Review And Reinforcement Chemistry Answers and Bestseller Lists
5. Accessing 14 3 Review And Reinforcement Chemistry Answers Free and Paid eBooks
  - 14 3 Review And Reinforcement Chemistry Answers Public Domain eBooks
  - 14 3 Review And Reinforcement Chemistry Answers eBook Subscription Services
  - 14 3 Review And Reinforcement Chemistry Answers Budget-Friendly Options

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  - ePub, PDF, MOBI, and More
  - 14 3 Review And Reinforcement Chemistry Answers Compatibility with Devices
  - 14 3 Review And Reinforcement Chemistry Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 14 3 Review And Reinforcement Chemistry Answers
  - Highlighting and Note-Taking 14 3 Review And Reinforcement Chemistry Answers
  - Interactive Elements 14 3 Review And Reinforcement Chemistry Answers
8. Staying Engaged with 14 3 Review And Reinforcement Chemistry Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 14 3 Review And Reinforcement Chemistry Answers
9. Balancing eBooks and Physical Books 14 3 Review And Reinforcement Chemistry Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 14 3 Review And Reinforcement Chemistry Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 14 3 Review And Reinforcement Chemistry Answers
  - Setting Reading Goals 14 3 Review And Reinforcement Chemistry Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 14 3 Review And Reinforcement Chemistry Answers
  - Fact-Checking eBook Content of 14 3 Review And Reinforcement Chemistry Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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