Name	Date	Class
14-1 Review and Reinford	cement (continued)	
Complete each of the sentences	below.	
15. The stronger the intermolec	rular forces in a liquid, the	its boiling point.
16. The state of a substance at r	room temperature depends on	
17. Water molecules in an ice of specifically.	cube are held together by bonds.	forces or, more
On the line at the left, write the le can be used once, more than on	tter of the term that matches each ce, or not at all.	description below. Each choice
a. dispersion	b. dipole-dipole c. hydrogo	en bonding
18. the force in between	een SO ₂ molecules	
19. the force that acc	ounts for HF being a liquid while	H ₂ and F ₂ are gases
20. the force that dep	the force that depends on temporary, induced dipoles	
21. the only intermol	ecular force that exists in noble ga	uses
	e between neighboring dipoles	
	es because of the large difference is	n electronegativities in the N-H
Answer the following question as	directed.	
 Which of the following state boiling point? Make a check 	ments could be used to explain w mark on the line for all that apply	hy water has an unusually high
a. Water molecules l	have a strong intermolecular force	the state of the s
	ogen atoms in water.	
c. Oxygen is more e	lectronegative than hydrogen.	
d. Water is an angul		
e. Oxygen atoms are	smaller than hydrogen atoms.	
	ntain metallic bonds.	
	fference in electronegativity betw	een oaveen and hydronen

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Edward P. Sarafino

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