#### THE STRUCTURE OF THE ATOMS

#### PAPER 2 : STRUCTURE

Question	Explanation	Marks
l(a)(i)	Diffusion Molecules	1
(11)		1.0
	<ul> <li>Made of tiny / discrete particles/ molecules</li> </ul>	35-02-03-03-0
(iiii)	- Move faster / rapidly	1+1+1
	- Between air / another particles	
(b)(i)	83 °C	10
(ii)	Remain, because heat absorbed is used to overcome the forces between particles.	1+1
Cititi)	Move faster	100
*********	Particles gain kinetic energy.	11
2(a)(i)	White fame	1
(11)	Ammonium chloride	1
(iiii)	Diffusion	1
(b)	Ammonia, because ammonia is lighter than hydrogen chloride.	1+1
(e)	NH <sub>3</sub> + HCl → NH <sub>4</sub> Cl	1
3(a)(i)	Number of proton in the nucleus of an atom.	400
(iii)	12	10
(b)	X and Y. Because they have same proton number but different neutron or nucleon number.	1+1
(c)	2.8.7	1
(d)(i)	( <b>©</b> )	1+1
(iii)	7	1
(iiii)	Y	L.
(e)	it X	Ti .
4(a0(ii)	Both axis are labeled with unit and consistent scales     All points transfer correctly	3.
	Size of graph at least 1% of the page of the graph paper     Curve of the graph is correct and smooth	
(iii)	- Size of graph at least 1/2 of the page of the graph paper	10
(ii) (iii)	<ul> <li>Size of graph at least ½ of the page of the graph paper</li> <li>Curve of the graph is correct and smooth</li> </ul>	1 2

# 450 12 M J 2010 Marking Scheme

Judd E. Hollander

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