



TENKASI DISTRICT Standard - 12

Time Allowed: 3.00 Hours

CHEMISTRY

Maximum Marks: 70

PART - I

Note: 1. Answer all the questions.

15×1=15

2. Choose the most appropriate answer from the given four alternatives and write the option code and the corresponding answer.

1. The metal oxide which cannot be reduced to metal by carbon is _____.
 a) PbO b) Al_2O_3 c) ZnO d) FeO
2. Match items in column - I with the items of column - II and assign the correct code.

Column - I

- A. Haematite
 B. Azurite
 C. Galena
 D. Calamine

Column - II

- (i) $2CuCO_3 \cdot Cu(OH)_2$
 (ii) $ZnCO_3$
 (iii) Fe_2O_3
 (iv) PbS

- | | A | B | C | D |
|----|-------|------|------|------|
| a) | (iii) | (iv) | (i) | (ii) |
| b) | (iii) | (ii) | (i) | (iv) |
| c) | (iii) | (iv) | (ii) | (i) |
| d) | (iii) | (i) | (iv) | (ii) |

3. Hybridisation of carbon in Graphite _____.
 a) sp^2 b) sp^3 c) sp^2d d) sp
4. Oxidation number of nitrogen in pernitrous acid ($HOONO$) _____.
 a) +3 b) +5 c) +1 d) +6
5. Inter halogen compounds of type AX_2 , about the structure of hybridization and bond pairs / lone pairs, which one is correct?
 a) T - shaped, sp^3d , $\frac{3}{2}$
 b) Square pyramidal, sp^3d^2 , $\frac{5}{1}$
 c) T - shaped, sp^3d , $\frac{5}{1}$
 d) Square pyramidal, sp^3d , $\frac{3}{2}$
6. Magnetic moment of Mn^{2+} is _____.
 a) 5.92 BM b) 8.95 BM c) 2.80 BM d) 3.90 BM
7. Number of electrons transferred to $KMnO_4$ in acid medium _____.
 a) 3 b) 1 c) 2 d) 5
8. An ionic compound A_xB_y crystallizes in fcc type crystal structure with B ions at the centre of each face and A ion occupying corners of the cube, the correct formula of A_xB_y is _____.
 a) AB b) AB_3
 c) A_3B d) A_8B_6
9. For the reaction, $2SO_2 + O_2 \rightarrow 2SO_3$, if $\frac{-d}{dt}[SO_2] = k_1[SO_2]$, $\frac{-d}{dt}[O_2] = k_2[SO_2]$, $\frac{-d}{dt}[SO_3] = k_3[SO_2]$ then the relation between k_1 , k_2 and k_3 is _____.
 a) $2k_1 = k_2 = k_3$ b) $k_1 = 2k_2 = k_3$
 c) $k_1 = k_2 = 2k_3$ d) $2k_1 = 2k_2 = k_3$
10. Which of the following can act as Lowry. Bronsted acid as well as base?
 a) HCl b) SO_4^{2-} c) HPO_4^{2-} d) Br

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