THE STRUCTURE OF THE ATOMS

PAPER 2 : STRUCTURE

Question	Explanation	Marks
l(a)(i)	Diffusion Molecules	1
(11)		10
	 Made of tiny / discrete particles/ molecules 	3.5-0.5-0.00
(iiii)	- Move faster / rapidly	1+1+1
	- Between air / another particles	
(b)(i)	83 °C	1.0
(ii)	Remain, because heat absorbed is used to overcome the forces between particles.	1+1
Cititi)	Move faster	100
********	Particles gain kinetic energy.	130
2(a)(i)	White fame	1
(10)	Ammonium chloride	1
(iiii)	Diffusion	1
(b)	Ammonia, because ammonia is lighter than hydrogen chloride.	1+1
(e)	NH ₂ + HCl → NH ₂ Cl	1.
3(a)(i)	Number of proton in the nucleus of an atom.	100
(iii)	12	10
(b)	X and Y. Because they have same proton number but different neutron or nucleon number.	1+1
(c)	2.8.7	1
(d)(i)	(O)	1+1
(iii)	7	10
CHIEF	Y	
Company Company		100
(e)	∦ x	1
(e)	Both axis are labeled with unit and consistent scales All points transfer correctly Size of graph at least ½ of the page of the graph paper Curve of the graph is correct and smooth	
(c) 4(a)(i)	Both axis are labeled with unit and consistent scales All points transfer correctly Size of graph at least ½ of the page of the graph paper	1
177.5	Both axis are labeled with unit and consistent scales All points transfer correctly Size of graph at least ½ of the page of the graph paper Curve of the graph is correct and smooth	3

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L Darling-Hammond

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