

DATA AND CALCULATIONS

A. pH of Acid Solutions:

1. Strong Acid	$[\text{HCl}]_1$	$[\text{HCl}]_2$
Measured pH	1.35	1.95
Molarity		
Calculated pH		
2. Weak Acid	$[\text{HC}_2\text{H}_3\text{O}_2]_1$	$[\text{HC}_2\text{H}_3\text{O}_2]_2$
Measured pH	2.84	3.20
Molarity		
Calculated pH		

Compare and explain the observed values of pH for these two acids.

Which acid pH changed the most on dilution? Explain why (consider the equilibrium occurring).

Ph Properties Of Buffer Solutions Lab Calculations

Hussin A.Rothana



Ph Properties Of Buffer Solutions Lab Calculations:

Whispering the Strategies of Language: An Psychological Quest through **Ph Properties Of Buffer Solutions Lab Calculations**

In a digitally-driven earth where monitors reign great and immediate connection drowns out the subtleties of language, the profound secrets and emotional subtleties hidden within words usually move unheard. However, set within the pages of **Ph Properties Of Buffer Solutions Lab Calculations** a captivating fictional treasure sporting with natural emotions, lies an exceptional journey waiting to be undertaken. Penned by an experienced wordsmith, that charming opus invites readers on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within the cloth of each word. Within the mental depths of the moving review, we will embark upon a honest exploration of the book is core themes, dissect their captivating publishing type, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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