

1-6, 8, 9, 11

Hydrolysis of Salts and pH of Buffer Solutions

Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

1. Define Brønsted-Lowry acids and bases.

According to Brønsted-Lowry definitions, an acid is a proton donor and a base is a proton acceptor.

2.

- Which of the following ions will react with water in a hydrolysis reaction:

K^+ , Ba^{2+} , Cu^{2+} , Zn^{2+} , F^- , SO_3^{2-} , and Cl^- ?

Cu^{2+} , Zn^{2+} , and SO_3^{2-} will undergo hydrolysis.

3. For those ions in question 2 that undergo hydrolysis, write net ionic equations for the hydrolysis reaction.

Example: $Cu^{2+}(aq) + 2H_2O(l) \rightleftharpoons Cu(OH)_2(s) + 2H^+(aq)$.

4. The K_a for HCN is 4.9×10^{-10} . What is the value of K_b for CN^- ?

$$K_b = \frac{1.0 \times 10^{-14}}{4.9 \times 10^{-10}} = 2.0 \times 10^{-5}$$

5. What are the conjugate base and conjugate acid of $H_2PO_4^-$?

The conjugate base is HPO_4^{2-} and the conjugate acid is H_3PO_4 .

6. From what acid and what base were the following salts made: K_2SO_4 , NH_4Cl , and $Ca(NO_3)_2$?

For $CaSO_4$, may be made from $Ca(OH)_2$ and H_2SO_4 .

7. Define the term salt.

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IM Harris



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Operative Dentistry, 2005

Gums and Stabilisers for the Food Industry 12 Glyn O Phillips, Peter A Williams, 2009-10-21 The latest volume in the successful Special Publication Series captures the most recent research findings in the field of food hydrocolloids The impressive list of contributions from international experts includes topics such as Hydrocolloids as dietary fibre The role of hydrocolloids in controlling the microstructure of foods The characterisation of hydrocolloids Rheological properties The influence of hydrocolloids on emulsion stability Low moisture systems Applications of hydrocolloids in food products Gums and Stabilisers for the Food Industry 12 with its wide breadth of coverage will be of great value to all who research produce process or use hydrocolloids both in industry and academia

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