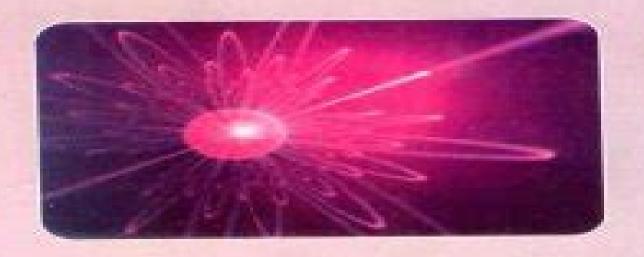
Huque and Nawab

Principles of Physical Chemistry



Fully Revised by

Dr. Muhammad Mahbubul Huque

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Chapter 5 Quantum Mechanics II Schrodinger wave equation for a particle in a three dimensional box The concept of degeneracy among energy levels for a particle in three dimensional box Schrodinger wave equation for a linear harmonic oscillator Zero point energy of a particle possessing harmonic motion and its consequence Schrodinger wave equation for three dimensional Rigid rotator Energy of rigid rotator Space quantization Schrodinger wave equation for hydrogen atom separation of variable in polar spherical coordinates and its solution Principle azimuthal and magnetic quantum numbers and the magnitude of their values Probability distribution function Radial distribution function Shape of atomic orbitals s p Law of mass action and its thermodynamic derivation Third law of thermodynamics Nernest heat theorem determination of absolute entropy unattainability of absolute zero and its limitation Phase diagram for two completely miscible components systems Eutectic systems Calculation of eutectic point Systems forming solid compounds Ax By with congruent and incongruent melting points Phase diagram and thermodynamic treatment of solid solutions Chapter 7 Chemical Dynamics II Chain reactions hydrogen bromine reaction pyrolysis of acetaldehyde decomposition of ethane Photochemical reactions hydrogen bromine General treatment of chain reactions ortho para hydrogen conversion and hydrogen bromine reactions Apparent activation energy of chain reactions Chain length Rice Herzfeld mechanism of organic molecules decomposition acetaldehyde Branching chain reactions and explosions H2 O2 reaction Kinetics of one intermediate enzymatic reaction Michaelis Menton treatment Evaluation of Michaelis s constant for enzyme substrate binding by Lineweaver Burk plot and Eadie Hofstae methods Competitive and non competitive inhibition Chapter 8 Electrochemistry II Ion Transport in Solutions Ionic movement under the influence of an electric field Mobility of ions Ionic drift velocity and its relation with current density Einstein relation between the absolute mobility and diffusion coefficient The Stokes Einstein relation The Nernst Einstein equation Walden's rule The Rate process approach to ionic migration. The Rate process equation for equivalent conductivity Total driving force for ionic transport Nernst Planck Flux equation Ionic drift and diffusion potential the Onsager phenomenological equations The basic equation for the diffusion Planck Henderson equation for the diffusion potential

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