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between the traditional solve a bunch of equations for a very simple model approach and the computational methods that are used to solve research problems This book will revolutionize the way physical chemistry is taught by bridging the gap between the traditional solve a bunch of equations for a very simple model approach and the computational methods that are used to solve research problems While some recent textbooks include exercises using pre packaged Hartree Fock DFT calculations this is largely limited to giving students a proverbial black box The DIY do it yourself approach taken in this book helps student gain understanding by building their own simulations from scratch The reader of this book should come away with the ability to apply and adapt these techniques in computational chemistry to his or her own research problems and have an enhanced ability to critically evaluate other computational results This book is mainly intended to be used in conjunction with an existing physical chemistry text such as McQuarrie Simon's Physical Chemistry A Molecular Approach but it is also well suited as a stand alone text for upper level undergraduate or intro graduate computational chemistry courses Assumes no computational background Enables students to build simulations from scratch to reproduce famous literature calculations Teaches a variety of computational numerical simulation methods applicable to solving chemical problems Designed to play well with McQuarrie Simon's landmark P CHEM text but can be used with others as well

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Molecular Physical Chemistry José J. C. Teixeira-Dias, 2017-01-16 This is the physical chemistry textbook for students with an affinity for computers It offers basic and advanced knowledge for students in the second year of chemistry masters studies and beyond In seven chapters the book presents thermodynamics chemical kinetics quantum mechanics and molecular structure including an introduction to quantum chemical calculations molecular symmetry and crystals The application of physical chemical knowledge and problem solving is demonstrated in a chapter on water treating both the water molecule as well as water in condensed phases Instead of a traditional textbook top down approach this book presents the subjects on the basis of examples exploring and running computer programs Mathematica discussing the results of molecular orbital calculations performed using Gaussian on small molecules and turning to suitable reference works to obtain thermodynamic data Selected Mathematica codes are explained at the end of each chapter and cross referenced with the text enabling students to plot functions solve equations fit data normalize probability functions manipulate matrices and test physical models In addition the book presents clear and step by step explanations and provides detailed and complete answers to all exercises In this way it creates an active learning environment that can prepare students for pursuing their own research projects further down the road Students who are not yet familiar with Mathematica or Gaussian will find a valuable introduction to computer based problem solving in the molecular sciences Other computer applications can alternatively be used For every chapter learning goals are clearly listed in the beginning so that readers can easily spot the **Problems and Solutions to** highlights and a glossary in the end of the chapter offers a quick look up of important terms Accompany McQuarrie and Simon's Physical Chemistry Heather Cox, 1998-02-12 This manual is designed to complement McQuarrie and Simon s new Physical Chemistry A Molecular Approach by providing a detailed solution for every one of the more than 1400 problems found in the text A Textbook of Physical Chemistry Arthur Adamson, 2012-12-02 A Textbook of Physical Chemistry Second Edition serves as an introductory text to physical chemistry Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry ideal and nonideal gases the three laws of thermodynamics thermochemistry and solutions of nonelectrolytes The kinetics of gas phase reactions colloids

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