

practice test ch 7 & 8

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following best describes the rates of chemical reaction?
- most chemical reactions occur very slowly
 - most chemical reactions occur at moderate rates
 - most chemical reactions occur very rapidly
 - chemical reactions have a wide range of rates, from extremely fast to extremely slow
- _____ 2. Which of the following is true of the rates of most chemical reactions?
- the initial rate is cannot be measured
 - the initial rate is faster than the rate later in time
 - the initial rate is the same as the rate later in time
 - the initial rate is slower than the rate later in time
- _____ 3. Which of the following is true of effective collisions?
- the number of effective collisions decreases as the temperature is increased
 - the number of effective collisions determines the reaction rate
 - both a. and b.
 - Neither a. nor b.
- _____ 4. Which of the following is true of effective collisions?
- the number of effective collisions determines the reaction rate
 - the number of effective collisions increases as the temperature is increased
 - both a. and b.
 - Neither a. nor b.
- _____ 5. Which of the following is true?
- decreasing the activation increases the reaction rate
 - the activation energy for an exothermic reaction is negative
 - both a. and b.
 - Neither a. nor b.
- _____ 6. In an energy diagram for a chemical reaction what species appears at the highest energy?
- the catalyst
 - the products
 - the reactants
 - the transition state
- _____ 7. Given that the reaction $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$ is exothermic, which of the following is true of the reaction $2 \text{H}_2\text{O}(\text{g}) \rightarrow 2 \text{H}_2(\text{g}) + \text{O}_2(\text{g})$?
- its activation energy is lower than that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is higher than that of c. its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - there is no relationship between its activation energy and that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
- _____ 8. In an energy diagram for an endothermic chemical reaction which of the following is true?
- the energy of the products is lower than that of the reactants
 - the energy of the transition state is higher than that of the reactants
 - both a. and b.
 - Neither a. nor b.
- _____ 9. Consider a reaction such as $\text{A}_2(\text{g}) + \text{B}_2(\text{g}) \rightarrow 2\text{AB}(\text{g})$. What can we say about the mechanism of the reaction?
- if the reaction is fast it occurs in a single step
 - if the reaction is fast it occurs in multiple steps
 - if the reaction is slow it occurs in multiple steps

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Elementary Modern Chemistry Wilhelm Ostwald, Harry Wheeler Morse, 1909 **Modern Electrochemistry 2B** John O'M. Bockris, Amulya K.N. Reddy, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States Modern NMR Techniques for Chemistry Research A.E. Derome, 2013-10-22 Presents an introduction to modern NMR methods at a level suited to organic and inorganic chemists engaged in the solution of structural and mechanistic problems The book assumes familiarity only with the simple use of proton and carbon spectra as sources of structural information and describes the advantages of pulse and Fourier transform spectroscopy which form the basis of all modern NMR experiments Discussion of key experiments is illustrated by numerous examples of the solutions to real problems The emphasis throughout is on the practical side of NMR and the book will be of great use to chemists engaged in both academic and industrial research who wish to realise the full possibilities of the new wave NMR *Introduction to Corrosion Science* Edward McCafferty, 2010-07-17 This textbook is intended for a one semester course in corrosion science at the graduate or advanced undergraduate level The approach is that of a physical chemist or materials scientist and the text is geared toward students of chemistry materials science and engineering This textbook should also be useful to practicing corrosion engineers or materials engineers who wish to enhance their understanding of the fundamental principles of corrosion science It is assumed that the student or reader does not have a background in electrochemistry However the student or reader should have taken at least an undergraduate course in materials science or physical chemistry More material is presented in the textbook than can be covered in a one semester course so the book is intended for both the classroom and as a source book for further use This book grew out of classroom lectures which the author presented between 1982 and the present while a

professorial lecturer at George Washington University Washington DC where he organized and taught a graduate course on Environmental Effects on Materials Additional material has been provided by over 30 years of experience in corrosion research largely at the Naval Research Laboratory Washington DC and also at the Bethlehem Steel Company Bethlehem PA and as a Robert A Welch Postdoctoral Fellow at the University of Texas The text emphasizes basic principles of corrosion science which underpin extensions to practice Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance **Modern Inorganic Chemistry** Joseph William Mellor, 1917 WAPD-BT, 1961 *Modern Chemistry* Nicholas D. Tzimopoulos, 1990 *Application of Solution Protein Chemistry to Biotechnology* Roger L. Lundblad, 2009-05-12 Reflecting the versatility of the author's science and the depth of his experience *Application of Solution Protein Chemistry to Biotechnology* explores key contributions that protein scientists can make in the development of products that are both important and commercially viable and provides them with tools and information required for successfu **Bettis Technical Review**, 1961-08 **General Chemistry** Ralph H. Petrucci, 1989 Books in Print Supplement, 2002 *Chemical Metallurgy* J.J. Moore, 2013-10-22 *Chemical Metallurgy* Second Edition provides the fundamental chemical principles and demonstrates the application of these principles to process metallurgy materials synthesis and processing and corrosion protection The book consists of nine chapters The first five chapters emphasize the fundamental chemical principles involved in metallurgical reactions An additional chapter on slag chemistry has also been added in this second edition in order to provide a more thorough understanding of slag metal reactions The final three chapters focus on the applications of the chemical principles to the extraction and refining of metals metal melting and recycling and metallic corrosion The book will be of value to materials students and teachers and scientists and engineers entering employment in the metallurgical and materials processing and metal finishing industries **Handbook of Modern Chemistry** Charles Meymott Tidy, 1878 **CRC Handbook of Basic Tables for Chemical Analysis** Thomas J. Bruno, Paris D.N. Svoronos, 2020-07-30 Researchers in chemistry chemical engineering pharmaceutical science forensics and environmental science make routine use of chemical analysis but the information these researchers need is often scattered in different sources and difficult to access The *CRC Handbook of Basic Tables for Chemical Analysis Data Driven Methods and*

Interpretation Fourth Edition is a one stop reference that presents updated data in a handy format specifically designed for use when reaching a decision point in designing an analysis or interpreting results This new edition offers expanded coverage of calibration and uncertainty and continues to include the critical information scientists rely on to perform accurate analysis Enhancements to the Fourth Edition Compiles a huge array of useful and important data into a single convenient source Explanatory text provides context for data and guidelines on applications Coalesces information from several different fields Provides information on the most useful wet chemistry methods as well as instrumental techniques with an expanded discussion of laboratory safety Contains information of historical importance necessary to interpret the literature and understand current methodology Unmatched in its coverage of the range of information scientists need in the lab this resource will be referred to again and again by practitioners who need quick easy access to the data that forms the basis for experimentation and analysis

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Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters, 1902

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Meyer,1979

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