

$$\text{mEq} = \frac{\text{mg}}{\text{gm Eq wt}}$$

To convert mg per 100 ml to mEq/litre

$$\text{mg per 100 ml} \times 10 = \text{mg/L}$$

So,
$$\frac{\text{mg per 100 ml} \times 10}{\text{Eq Wt}} = \text{mEq/L}$$

Since
$$\text{Eq Wt} = \frac{\text{molecular weight}}{\text{valency}}$$

$$\frac{\frac{\text{mg/100 ml} \times 10}{\text{molecular weight}}}{\text{valency}} = \text{mEq/L}$$

Or
$$\frac{\text{mg/100 ml} \times 10 \times \text{valency}}{\text{molecular weight}} = \text{mEq/L} \quad \dots$$

Modern Chemistry Chapter 5 Mixed Review Answers

JR Anderson



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Encyclopaedia Metropolitana: Plates to Mixed Sciences, Vol. 5 and 6** Edward Smedley, Hugh James Rose, Henry John Rose, 1845 Modern Machinery, 1907 **I & EC**, 1925 *Industrial & Engineering Chemistry*, 1925 **Saturday Review**, 1876 *Modern Electrochemistry 2B* John O'M. Bockris, Amulya K.N. Reddy, 2007-05-08 This book had its nucleus in some lectures given by one of us J O M B in a course on electrochemistry to students of energy conversion at the University of Pennsylvania It was there that he met a number of people trained in chemistry physics biology metallurgy and materials science all of whom wanted to know something about electrochemistry The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered The lectures were recorded and written up by Dr Klaus Muller as a 293 page manuscript At a later stage A K N R joined the effort it was decided to make a fresh start and to write a much more comprehensive text Of methods for direct energy conversion the electrochemical one is the most advanced and seems the most likely to become of considerable practical importance Thus conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met Corrosion is recognized as having an electrochemical basis The synthesis of nylon now contains an important electrochemical stage Some central biological mechanisms have been shown to take place by means of electrochemical reactions A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States **English Mechanic and Mirror of Science and Art**, 1887 **English Mechanic and Mirror of Science**, 1887 **New Technical Books** New York Public Library, 1993 **American Druggist and Pharmaceutical Record**, 1895 **Engineering and Mining Journal**, 1903 **The Publishers' Circular and Booksellers' Record**, 1915 **English Mechanic and World of Science**, 1887 The Floricultural Cabinet, and Florists Magazine, 1911 Scientific American, 1898 **Practical Chemistry** Newton Henry Black, James Bryant Conant, 1920 **Journal of Gas Lighting**, 1917 Book Review Digest, 1996-08 **The Gardeners' Chronicle and Agricultural Gazette**, 1857

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