CHAPTER 8 REVIEW

Chemical Equations and Reactions

SECTION 8-1

SHORT ANSWER Answer the following questions in the space provided.

1. Match the symbol on the left with its appropriate description on the right.

_____ A

(a) A precipitate forms.

(b) A gas forms.

_____ ↑

(c) A reversible reaction occurs.

....(1)

(d) Heat is applied to the reactants.

____ (aq)

(e) A chemical is dissolved in water.
(f) A chemical is in the liquid state.

2. Finish balancing the following equation:

 $3Fe_3O_4 + ____ Al \rightarrow ___ Al_2O_3 + ____ Fe$

3. In each of the following formulas with coefficients, write the total number of atoms present.

_____ a. 4SO₂

_____ b. SO₂

_____ c. 3Al₂(SO₄)₃

_____ d, 6×10^{23} HNO₃

- Convert the following word equation into a balanced chemical equation: aluminum metal + copper(II) fluoride → aluminum fluoride + copper metal
- 5. One way to test the salinity of a water supply is to add a few drops of silver nitrate of known concentration to the water. As the solutions of sodium chloride and silver nitrate mix, a precipitate of silver chloride forms, leaving sodium nitrate in solution. Translate these sentences into a balanced chemical equation.
- a. Balance the following equation: NaHCO_y(s)
 [∆]⇒ Na₂CO_y(s) + H₂O(g) + CO₂(g)

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Modern Chemistry Chapter 8 Review Answer Key

Leo J. Malone, Theodore O. Dolter

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

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