

Chemistry Chapter 5 Exam

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
 - a. Mendeleev.
 - b. Moseley.
 - c. Bohr.
 - d. Ramsay.
- _____ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
 - a. atomic numbers.
 - b. colors.
 - c. properties.
 - d. radioactivity.
- _____ 3. Mendeleev attempted to organize the chemical elements based on their
 - a. symbols.
 - b. properties.
 - c. atomic numbers.
 - d. electron configurations.
- _____ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
 - a. atomic number.
 - b. density.
 - c. reactivity.
 - d. atomic mass.
- _____ 5. Mendeleev is credited with developing the first successful
 - a. periodic table.
 - b. method for determining atomic number.
 - c. test for radioactivity.
 - d. use of X rays.
- _____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
 - a. properties.
 - b. atomic numbers.
 - c. densities.
 - d. colors.
- _____ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
 - a. atomic number.
 - b. electron configuration.
 - c. isotopes.
 - d. properties.
- _____ 8. Mendeleev predicted that the spaces in his periodic table represented
 - a. isotopes.
 - b. radioactive elements.
 - c. permanent gaps.
 - d. undiscovered elements.
- _____ 9. Mendeleev's table was called periodic because the properties of the elements
 - a. showed no pattern.
 - b. occurred at repeated intervals called periods.
 - c. occurred at regular time intervals called periods.
 - d. were identical.
- _____ 10. The person whose work led to a periodic table based on increasing atomic number was
 - a. Moseley.
 - b. Mendeleev.
 - c. Rutherford.
 - d. Cannizzaro.
- _____ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
 - a. atomic mass.
 - b. density.
 - c. radioactivity.
 - d. atomic number.
- _____ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
 - a. Meyer
 - c. Stas

Modern Chemistry Chapter 5 Test

Harold Clark Metcalfe



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Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters,1902 **Prentice Hall**
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Organic Chemistry Carl H. Snyder,1973 **Modern Methods of Drug Discovery** Alexander Hillisch,Rolf

Hilgenfeld,2002-12-11 Research in the pharmaceutical industry today is in many respects quite different from what it used to be only fifteen years ago There have been dramatic changes in approaches for identifying new chemical entities with a desired biological activity While chemical modification of existing leads was the most important approach in the 1970s and 1980s high throughput screening and structure based design are now major players among a multitude of methods used in drug discovery Quite often companies favor one of these relatively new approaches over the other e g screening over rational design or vice versa but we believe that an intelligent and concerted use of several or all methods currently available to drug discovery will be more successful in the medium term What has changed most significantly in the past few years is the time available for identifying new chemical entities Because of the high costs of drug discovery projects pressure for maximum success in the shortest possible time is higher than ever In addition the multidisciplinary character of the field is much more pronounced today than it used to be As a consequence researchers and project managers in the pharmaceutical industry should have a solid knowledge of the more important methods available to drug discovery because it is the rapidly and intelligently combined use of these which will determine the success or failure of preclinical projects **Equilibria in**

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