

DU MSc Chemistry

Topic:- CHEM MSC 52

1) The oxo-acid of phosphorus having P-atoms in +4, +3, and +4 oxidation states respectively, is
[Question ID = 9737]

1. $\text{H}_3\text{P}_2\text{O}_6$ [Option ID = 38942]
2. $\text{H}_5\text{P}_2\text{O}_7$ [Option ID = 38944]
3. $\text{H}_5\text{P}_2\text{O}_8$ [Option ID = 38946]
4. $\text{H}_5\text{P}_2\text{O}_9$ [Option ID = 38948]

Correct Answer :-

- $\text{H}_5\text{P}_2\text{O}_8$ [Option ID = 38946]

2) According to Wade's theory the anion $[\text{B}_{10}\text{H}_{12}]^{2-}$ adopts
[Question ID = 9740]

1. closo-structure [Option ID = 38951]
2. arachno-structure [Option ID = 38953]
3. nido-structure [Option ID = 38955]
4. mido-structure [Option ID = 38957]

Correct Answer :-

- closo-structure [Option ID = 38951]

3) Which one of the following species has the magnetic moment value of 3.87 BM?
[Question ID = 9741]

1. Fe^{2+} [Option ID = 38958]
2. Cr^{2+} [Option ID = 38960]
3. Ce^{2+} [Option ID = 38962]
4. Au^{2+} [Option ID = 38964]

Correct Answer :-

- Ce^{2+} [Option ID = 38962]

4) Which of the following complexes does not show optical activity?
[Question ID = 9743]

1. $[\text{Co}(\text{EDTA})]^-$ [Option ID = 38966]
2. $[\text{Pt}(\text{en})_2]^{2+}$ [Option ID = 38968]
3. $[\text{Pt}(\text{pn})_2]^{2+}$ [Option ID = 38970]
4. $[\text{Pt}(\text{en})_2]^{2+}$ [Option ID = 38972]

Correct Answer :-

- $[\text{Pt}(\text{en})_2]^{2+}$ [Option ID = 38972]

5) The expected H-H-H bond angle in $[\text{H}_3]^+$
[Question ID = 9745]

1. 180° [Option ID = 38974]
2. 135° [Option ID = 38976]
3. 60° [Option ID = 38978]
4. 90° [Option ID = 38980]

Correct Answer :-

- 60° [Option ID = 38978]

6) The metallic radii are abnormally high for which of the following pairs?
[Question ID = 9747]

1. Eu, Yb [Option ID = 38982]
2. Sm, Tm [Option ID = 38984]
3. Gd, Lu [Option ID = 38986]
4. Nd, Ho [Option ID = 38988]

Correct Answer :-

- Eu, Yb [Option ID = 38982]

7) The least basic among the following is
[Question ID = 9750]

1. $\text{Al}(\text{OH})_3$ [Option ID = 38991]
2. $\text{La}(\text{OH})_3$ [Option ID = 38993]
3. $\text{Ce}(\text{OH})_3$ [Option ID = 38995]
4. $\text{Lu}(\text{OH})_3$ [Option ID = 38997]

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