

Experiment 11A: The Periodic Table

Procedure

Part I Electronic Configuration and the Periodic Table

1. Attach a separate sheet of paper.

Part II Atomic Number, Atomic Number, Subatomic Particles and the Periodic Table

1. Attach a separate sheet of paper.
2. Attach a 2 x 20 cm

Part III Atomic Configuration and the Periodic Table

1. Attach a separate sheet of paper.

Objectives

1. Explain the phenomenon of electron configuration (Figure 11.1.1). What relationship can be seen between an element's electron configuration and its atomic configuration?

Students will understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table.

2. Explain the graph of atomic number plotted against increasing atomic number (Figure 11.1.2). What is particularly noteworthy about it? Is it, describing the periodic table?

Students will understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table.

3. Explain the graph of atomic number plotted against increasing atomic number (Figure 11.1.3). What is particularly noteworthy about it? Is it, describing the periodic table?

Students will understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table.

4. Explain the graph of atomic number plotted against increasing atomic number (Figure 11.1.4). What is particularly noteworthy about it? Is it, describing the periodic table?

Students will understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table. They will also understand the relationship between the atomic number and the periodic table.

Laboratory Experiment 11a Answers Periodic Table

Lauren Gardner



Laboratory Experiment 11a Answers Periodic Table:

Elements and the Periodic Table, Grades 5 - 12 Theodore A. Abbgy, 2001-01-01 This informative classroom supplement is a great introduction to the periodic table explored in sequential form It includes activities transparency masters a teacher s guide an element game quizzes tests rubrics and answer keys Unit topics include discovering what elements are the uses of the elements element symbols periodic table organization and more Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources Experimental Chemistry Robert J. Artz, 1982 **Principles Of Descriptive Inorganic**

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