

CHAPTER 7

7.1) DENSITY OF ORBITALS IN ONE AND TWO DIMENSIONS

(a) In one dimension, $\psi(x) = A \sin(n\pi x/L)$, analogously to (3.58). Here n is an integer. The number of electrons that can be accommodated in the orbitals up to a given quantum number n is $N = 2n$, where the factor 2 arises from the electron spin. The corresponding maximum energy is

$$\epsilon = \frac{\hbar^2}{2m} \left(\frac{n\pi}{L} \right)^2 = \frac{1}{2m} \left(\frac{\pi \hbar N}{2L} \right)^2 ; N = (2L/\pi \hbar) (2m\epsilon)^{1/2} . \quad (S1)$$

If N is the total number of electrons actually present, then ϵ is the Fermi energy ϵ_F . From (S1):

$$\frac{d\epsilon}{\epsilon} = \frac{2dN}{N} , \quad \mathcal{D}_1(\epsilon) = \frac{dN}{d\epsilon} = \frac{N}{2\epsilon} = (L/\pi \hbar) (2m/\epsilon)^{1/2} .$$

(b) In two dimensions, the relation between the number of electrons N and the highest orbital n is

$$N = 2 \times (\pi n^2/4) , \quad n = (2N/\pi)^{1/2} .$$

Therefore

$$\epsilon = \frac{\hbar^2}{2m} \left(\frac{n\pi}{L} \right)^2 = \frac{\pi \hbar^2 N}{mL^2} , \quad N = L^2 m \epsilon / \pi \hbar^2 ,$$

$$\mathcal{D}_2(\epsilon) = \frac{dN}{d\epsilon} = L^2 m / \pi \hbar^2 = Am / \pi \hbar^2 .$$

7.2) ENERGY OF RELATIVISTIC FERMI GAS

(a) With the help of (6) -- $n_F = (3N/\pi)^{1/3}$, which remains valid -- we obtain, with $n = N/L^3$:

$$\epsilon_F = p_F c = (\hbar \pi c / L) n_F = \hbar \pi c (3n/\pi)^{1/3} .$$

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Ernest B. Yeager, Alvin J. Salkind



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Berkeley Physics Course: Mechanics, by C. Kittel, W. D. Knight, and M. A. Ruderman Charles Kittel, 1965 College physics course for students majoring in science and engineering

An Introduction To Statistical Thermodynamics Robert P H Gasser, Richards W Graham, 1995-09-26 Statistical thermodynamics plays a vital linking role between quantum theory and chemical thermodynamics yet students often find the subject unpalatable In this updated version of a popular text the authors overcome this by emphasising the concepts involved in particular demystifying the partition function They do not get bogged down in the mathematical niceties that are essential for a profound study of the subject but which can confuse the beginner Strong emphasis is placed on the physical basis of statistical thermodynamics and the relations with experiment After a clear exposition of the distribution laws partition functions heat capacities chemical equilibria and kinetics the subject is further illuminated by a discussion of low temperature phenomena and spectroscopy The coverage is brought right up to date with a chapter on computer simulation and a final section which ranges beyond the narrow limits usually associated with student texts to emphasise the common dependence of macroscopic behaviour on the properties of constituent atoms and molecules Since first published in 1974 as Entropy and Energy Levels the book has been very popular with students This revised and updated version will no doubt serve the same needs

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Sclerotherapy E-Book Mitchel P. Goldman, Robert A Weiss, 2016-09-19 This classic resource by Drs Mitchel P Goldman Robert A Weiss and Jean Jerome Guex provides highly practical up to date guidance for the effective management of varicose veins and other vascular anomalies It is an indispensable reference for a wide audience including dermatologists invasive radiologists family practitioners vascular and cosmetic surgeons Clearly written by global experts Sclerotherapy 6th Edition helps those new to the field to gain a firm understanding of successful techniques as well as showing seasoned practitioners how to improve and hone their skills with today's best and newest approaches Case studies and detailed color illustrations offer step by step visual guidance Covers everything you need to know with a practical approach from the pathogenesis of varicosities to diagnostic and treatment options including evidence based decision making Helps you optimize outcomes and improve your surgical injection and laser

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Entropy and Energy Levels Robert Paul Holland Gasser, William Graham Richards, 1974 *Techniques of Electrochemistry* Ernest B. Yeager, Alvin J. Salkind, 1972 **Traité de chimie analytique minérale, qualitative et quantitative** Lucien Louis de Koninck, 1918 **Ordering and Phase Transitions in Charged Colloids** Akhilesh Kumar Arora, Butchi Venkata Rao Tata, 1996 Physical chemists will find this book comprehensive Topical reviews on all aspects of colloidal ordering and related phase transitions will be covered It provides a good blend of experimental and theoretical investigations Useful to materials scientists and chemical engineers the book includes a discussion of stability important from the point of view of applications of colloidal crystals Physicists will be interested in the book because it highlights the controversy over effective interparticle interaction in charged colloids

Photographic Science and Engineering , 1981 Sclerotherapy Mitchel P. Goldman, John J. Bergan, 2007 This edition provides the comprehensive coverage of all aspects of sclerotherapy and surgical treatment of varicose and telangiectatic leg veins It has been completely revised with all figures and drawings now in full color It provides extensive discussions of the latest techniques solutions and possible complications *Varicose Veins, Related Diseases and Sclerotherapy* H. I. Biegeleisen, 1984 Chekhoslovat'skiĭ Fizicheskiĭ Zhurnal. Czechoslovak Journal of Physics , 1980-07 **Fundamentals of Crystal Physics** И. У. Си́рови́ч Isakovitch Sirotin, Marianna Petrovna Shaskol'skai'a, 1982 **Crystal Physics** German Stepanovich Zhdanov, 1965

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web one unknown in each sample is an organic acid e g carboxylic acid phenol or sulfonic acid the other unknown in each sample is a neutral organic compound e g ketone ester amide etc the unknowns will be separated by extraction monitored by thin layer chromatography

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web 1 you have an unknown that is either an alcohol or a phenol using the following 1h nmr data is the unknown an alcohol or a phenol briefly explain your answer observed resonances doublet at 1 3 ppm septet at 2 8 ppm doublet at 3 7 ppm and a singlet at 4 1 ppm 2 you have an unknown organic compound that is either a ketone or an

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web the chromatography for gc and gc ms is identical in theory however the column used in the gc ms experiment is a capillary column as opposed to the packed column used in the gc experiment done in chemistry 105 a capillary column is simply a long tube made of glass with a small internal diameter for this experiment a 30 cm column with an

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web introduction along with synthesis and the examination of the reaction mechanisms an equally important part of organic chemistry has to do with the characterization and identification of compounds which may be encountered in sources ranging from a laboratory reaction to exotic tropical plants

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web nov 7 2021 lab report on identification of unknown organic compounds course chemistry institution bicol university the purpose of this laboratory activity was to identify the functional groups present in the given unknown and eventually identify what organic compound the unknown is preview 2 out of 5 pages report copyright violation other

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web the title of the abstract is the same as the title of the longer report that it summarizes the authors name and the place where the work was done are included using melting point to determine the identity of an unknown organic acid martha a hass albany college of pharmacy organic chemistry lab tuesday morning section june 15 2002 the

example unknowns lab report organic chemistry lab final - Oct 29 2022

web identification of unknowns marks a point of mastery within an organic chemistry lab because it involves the unification of a series of systematic tests observations and analysis that only when considered all together can provide the complete picture of

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web pre lab report 5 identification of unknown organic compounds pre lab grading rubric name rachel studocu points are assigned 5 pts total score 10 pts the liquid to its original bottle do not put other compounds or contaminated unknowns back into 2

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web apr 12 2020 all prelabs and lab reports are graded on a 100 point scale with the following distribution individual prelab 100 pts due at the beginning of each experiment

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web identification of unknown o attach ir and nmr with relevant peaks labeled things listed above it was determined that the unknown molecule was piperonylamine and when looking up the molecule the molecule also had a melting point of 138 139 c and has a ama of 151

experiment 7 9 lab report identification of unknown organic studocu - Jul 06 2023

web chem 3306 lab report n1 lab report 5 lab related documents experiment 1 exp1 3106 313 exp 4 lab report quiz 12 quiz 7 quiz 3 quiz 2 preview text identification of unknown organic compound purpose the purpose of this lab is to collect qualitative data on an unknown compound in order to discover the identity of the unknown via

24 qualitative organic analysis identification of an unknown - Aug 07 2023

web qualitative analysis of an organic unknown is an important proficiency for any student completing the organic chemistry laboratory course this experiment culminates the laboratory techniques that you have learned this semester and requires you to put the various pieces of information together to identify an unknown organic molecule

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web organic chemistry ii laboratory chem 2081 formal lab report lab ochem 2 lab report 7 identification of an unknown a2 taylor breech november 30 2016 chem nirmal mitra introduction the purpose of this lab experiment was to use three

different

8 identification of unknowns experiment chemistry libretexts - Sep 08 2023

web unknown reports you should submit an unknown preliminary report as soon as you have completed the experiments described below for the solid unknown you will take an accurate mp an ir spectrum using the kbr procedure solubility tests and based upon an analysis of this information do specific functionality tests as described in the manual

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web ms the spectrum for the unknown shows the molecular ion at 135.99 m/z the base peak is at 56.98 m/z which is the peak identified to be a butane chain of carbons the m/z peak is at 138.99 m/z and is roughly the same size as the m peak indicating the presence of bromine in this compound

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web in this paper several methods for identification characterization tests of organic compounds determination of molecular weight solubility of unknown compounds tlc technique each of you will receive two vials one containing a liquid unknown and one a solid unknown

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web 13 organic chemistry post lab 6 nutmeg and preparation of myristic acid organic chemistry laboratory i chm2210l assignments 100/17 organic chemistry lab report 12 organic chemistry laboratory i chm2210l assignments 100/17 experiment 6 isolation of trinyristin from nutmeg and preparation of myristic acid from

orgo 2 lab 6 lab report lab 6 an unknown binary mixture studocu - Feb 01 2023

web this lab report is based upon unknown binary mixture e containing unknowns 6s and 15l as it contains a solid liquid mixture vacuum filtration was conducted 6s is insoluble in 15l procedure 1 take the unknown binary mixture and make observations about the mixture by stating its state odor color and solubility characteristics