

5

ATOMIC STRUCTURE AND THE PERIODIC TABLE

Conceptual Curriculum

- Concrete concepts
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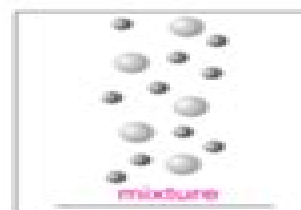
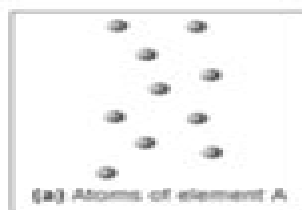
- ◆ Core honors content
- ◇ Options to accelerate

SECTION 5.1 ATOMS (pages 107–108) ● ■ ◇

This section describes early atomic theories of matter and provides ways to understand the tiny size of individual atoms.

► Early Models of the Atom (pages 107–108)

- Democritus of Abdera, who lived in Greece during the fourth century B.C., suggested that matter is made up of tiny particles that cannot be divided. He called these particles atoms.
- List two reasons why the ideas of Democritus were not useful in a scientific sense. They did not explain chemical behavior, and they lacked experimental support because scientific testing was unknown at the time.
- The modern process of discovery about atoms began with the theories of an English schoolteacher named John Dalton.
- Circle the letter of each sentence that is true about Dalton's atomic theory.
 - ☒ All elements are composed of tiny, indivisible particles called atoms.
 - ☐ An element is composed of several types of atoms.
 - ☒ Atoms of different elements can physically mix together, or can chemically combine in simple, whole-number ratios to form compounds.
 - ☒ Chemical reactions occur when atoms are separated, joined, or rearranged; however, atoms of one element are never changed into atoms of another element by a chemical reaction.
- In the diagram, use the labels *mixture* and *compound* to identify the mixture of elements A and B and the compound that forms when the atoms of elements A and B combine chemically.



Atomic Structure Periodic Table Answer Key

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teach the subjects beyond preparation for examinations what they are intending to achieve with the curriculum how well it is planned and enacted in classrooms and how they know whether it is doing what it is supposed to Fundamental to this understanding are the conversations between subject leaders and their line managers However there is sometimes a mismatch between the subject specialisms of senior leaders and those they line manage If I don't know the terrain and the importance of a particular subject how can I talk intelligently with colleagues who are specialists This book sets out to offer some tentative answers to these questions Each of the national curriculum subjects is discussed with a subject leader and provides an insight into what they view as the importance of the subject how they go about ensuring that knowledge understanding and skills are developed over time how they talk about the quality of the schemes in their departments and what they would welcome from senior leaders by way of support We have chosen this way of opening up the potentially difficult terrain of expertise on one side and relative lack of expertise on the other by providing these case studies They are suggested as prompts rather than the last word Informed debate is after all the fuel of curriculum development And why Huh Well Huh may be John's first response when he walks into a Year 8 German class but in fact we chose Huh as the title of our book as he is the Egyptian god of endlessness As Claire Hill so eloquently comments in her chapter Curriculum development is an ongoing process it is not going to be finished ever And we believe that Huh captures a healthy and expansive way of considering curriculum conversations

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McClendon, Grace Medes, 1925 **Report for the Year ...** National Physical Laboratory (Great Britain), 1925 Vol for 1905 include lists of papers published by the laboratory or communicated by members of the staff to scientific societies or to the technical journals Supplementary Report and Scheme of Work for the Year 1919-1920 National Physical Laboratory (Great Britain), 1924 Vols for 1905 51 include lists of reports and papers published by the laboratory **The Chemical News and Journal of Industrial Science**, 1921 Some Physico-chemical Themes Alfred Walter Stewart, 1922 **The University of Michigan Bulletin** University of Michigan, 1975 Each number is the catalogue of a specific school or college of the University **The Structure of the Atom** Edward Neville da Costa Andrade, 1924 **The Colorado School of Mines Magazine**, 1928 **The Chemical News and Journal of Physical Science**, 1918 Chemical News and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1919 **Chemical News and Journal of Industrial Science**, 1918

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