

Q.6. What is the particle in a nucleus that determines the identity of the element?

- a) Mass number
- b) Electron
- c) Proton
- d) Neutron

Q.7. The number of _____ can be calculated by subtracting its atomic number from its mass number.

- a) photons
- b) electrons
- c) protons
- d) neutrons

Q.8. Complete the following chart.

Symbol	Element	Atomic Number	Number of Protons	Number of Neutrons	Mass Number
$^{17}_8\text{O}$					
$^{18}_8\text{O}$					
$^{63}_{27}\text{Co}$					
$^{222}_{88}\text{Ra}$					
$^{226}_{88}\text{Ra}$					

Atomic Mass And Number Worksheet Answers

Neena Sinha, Anita Marwah



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