

POGIL Activity – Gas Variables

(Teacher Notes)

- Assessment Questions and Responses:

Assessment Questions

1. Select the equation below that gives the correct relationship between the internal pressure of a gas and the volume of a flexible container.
 - a. $P = k/r$
 - b. $P = kV$
 - c. $V = kP$
 - d. $k = p/V$
2. Explain, on the molecular level, the change in internal pressure of a gas sample in a nonflexible container when the temperature is raised.
3. Explain, on the molecular level, the change in volume in a flexible container when more gas is added.

Assessment Target Responses

1. a.
2. *When the temperature of a gas is increased, the average kinetic energy of the molecules increases. This makes the gas molecules hit harder and more often, which increases the pressure.*

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