

QUESTION

Determine the boiling point of ethylene glycol to water solution of different composition

STEP 1

1. Determine the boiling point of pure water: The boiling point of pure water is 100°C at standard atmospheric pressure.

STEP 2

2. Determine the boiling point of pure ethylene glycol: The boiling point of pure ethylene glycol is 197.3°C at standard atmospheric pressure.

Determine The Boiling Point Of Ethylene Glycol Water Solution Of Different Composition

A Loxley



Determine The Boiling Point Of Ethylene Glycol Water Solution Of Different Composition:

Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant *MTG CBSE Class 12 Chapterwise Question Bank Chemistry (For 2024 Exams)* MTG Learning Media,

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Raymond Chang, Charles M. Lovett, 2025-02-25 *Physical Chemistry for the Biosciences* has been optimized for a one semester course in physical chemistry for students of biosciences or a course in biophysical chemistry Most students enrolled in this course have taken general chemistry organic chemistry and a year of physics and calculus Fondly known as Baby Chang this best selling text is back in an updated second edition for the one semester physical chemistry course Carefully crafted to match the needs and interests of students majoring in the life sciences *Physical Chemistry for the Biosciences* has been revised to provide students with a sophisticated appreciation for physical chemistry as the basis for a variety of interesting biological phenomena Major changes to the new edition include Discussion of intermolecular forces in chapter Detailed discussion of protein and nucleic acid structure providing students with the background needed to fully understand the biological applications of thermodynamics and kinetics described later in the book Expanded and updated descriptions of biological examples such as protein misfolding diseases photosynthesis and vision **Advanced**

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