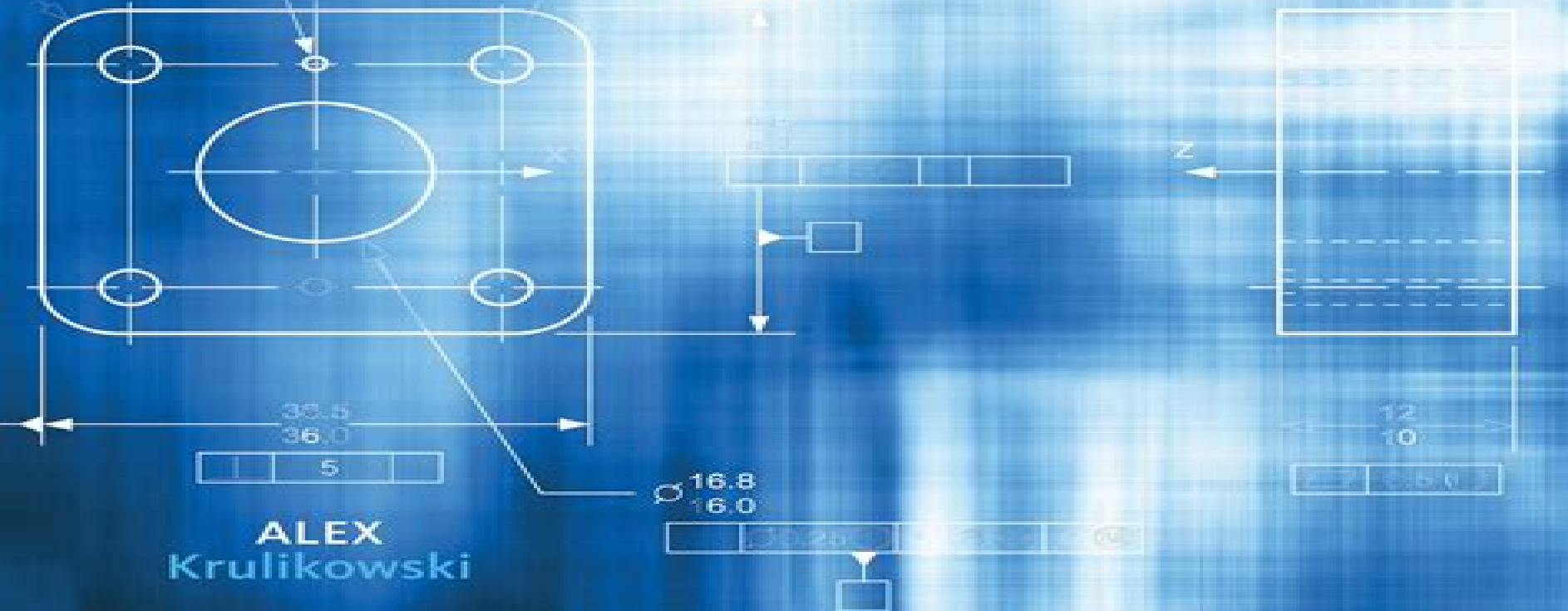


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web he proposed the atomic theory in 1803 which stated that 1 all matter was composed of small indivisible particles termed atoms 2 atoms of a given element possess unique characteristics and weight and 3 three types of atoms exist simple elements compound simple molecules and complex complex molecules

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web dalton introduced atomic theory by explaining the law of constancy of chemical composition and formulating the laws of partial pressures and multiple proportions he also carried out numerous tests on gas absorption in liquids and thermal expansion of gases and described the condition known as daltonism

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web dalton s atomic theory 1804 from his own experiments and observations as well as the work of his peers dalton proposed a new theory of the atom this later became known as dalton s atomic theory the general tenets of this theory are as follows all matter is composed of extremely small particles called atoms

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web science history institute although a schoolteacher a meteorologist and an expert on color blindness john dalton is best known for his pioneering theory of atomism he also developed methods to calculate atomic weights and structures and formulated the law of partial pressures

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web however the essential novelty of dalton s atomic theory is that he provided a method of calculating relative atomic weights for the chemical elements which provides the means for the assignment of molecular formulas for all chemical substances

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