(Solution Manual all Chapters)

#### SOLUTIONS CHAPTER 1

#### 1.1. Show that Equation (1.6) follows from Equation (1.3).

Solution: Equation (1.3) is  $dE_p = \frac{q^2}{4\pi c_r r^2} dr$ . Integrating both sides we obtain

$$\int dE_{r} = \int \frac{q^{2}}{4\pi\epsilon_{0}r^{2}} dr = \frac{q^{2}}{4\pi\epsilon_{0}} \int \frac{1}{r^{2}} dr = \frac{q^{2}}{4\pi\epsilon_{0}} \left[ -\frac{1}{r} \right] + const = E_{r}$$

To find the constant, we employ the boundary condition that at  $r = \infty$ ,  $E_P = E_{\text{tot}}$ :

$$E_F(r = \infty) = E_{\infty} = \frac{-q^2}{4\pi\varepsilon_0 \cdot \infty} + const = 0 + const$$

$$z$$
,  $const = E_{rac}$  and

$$E_{\mu} = E_{\text{usr}} - \frac{-q^2}{4\pi\epsilon_{\text{N}}\cdot\infty} = E_{\text{usr}}$$
, Equation (1.6)

## 1.2. Consider a lithium nucleus, of charge +3q. Calculate the first three electron energies for an electron in a Li\*\* ion, using the Bohr model.

We repeat the analysis that we used for the hydrogen atom, except that now the charge of the nucleus  $Q_1$  is equal to  $3q = +3(1.6 \times 10^{-19})C$ . The results of the key steps are

$$F = \frac{Q_1Q_2}{4\pi\epsilon_0 r^2} = \frac{-3q^2}{4\pi\epsilon_0 r^2}$$

$$E_p(r) = E_{ne} - \frac{3q^2}{4\pi\epsilon_n r}$$

$$\frac{mv^2}{r} - \frac{3q^2}{4\pi\epsilon_0 r^2} = 0$$

$$mev_n r_n = mh$$

$$\nu_n = \frac{3q^2}{(4\pi\epsilon_0)\hbar} \left(\frac{1}{n}\right)$$

$$r_{s} = \frac{\left(4\pi\varepsilon_{o}\right)\hbar^{2}}{3m\sigma^{2}}(n^{2})$$

$$E_K = \frac{m3^2 q^4}{2(4\pi\epsilon_0)^2 \hbar^2} \left(\frac{1}{n^2}\right)$$

$$E_{\pi} = E_{p_{\pi}} + E_{Ee} = E_{ee} - \frac{9mq^4}{2(4\pi\epsilon_0)^2 \hbar^2} \left(\frac{1}{n^2}\right)$$

Thus  $E_i = E_{inv} - 9(13.6eV) = E_{inv} - 122eV$ 

# <u>Fundamentals Of Semiconductor Devices Anderson</u> <u>Solution Manual</u>

**Robert F. Pierret** 

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