Gas Variables

How are the variables that describe a gas related?

Why?

Imagine buying a balloon bouquet at a party store. How will the helium gas in the bouquet behave if you carry it outside on a hot summer day? How will it behave if you carry it outside during a snowstorm? What happens if the balloons are made of latex, which can stretch? What happens if the balloons are made of Mylar®, which cannot stretch? What if you add just a small amount of gas to each balloon? What if you add a lot of gas? In this activity, you will explore four variables that quantify gases—pressure (P), volume (V), temperature (T), and moles (n) of gas. These four variables can be related mathematically so that predictions about gas behavior can be made.

Model 1 – Gases in a Nonflexible Container

Experiment A (Adding more gas)



Volume - 1 unit External pressure = 1 atm External pressure = 1 atm External pressure = 1 atm. Internal pressure = 1 atm Internal pressure = 2 atm Internal pressure = 3 atm Temperature = 200 K



Volume - 1 unit Temperature - 200 K



Volume = 1 unit Temperature = 200 K

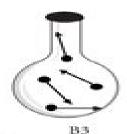
Experiment B (Heating the gas)



Volume - 1 unit External pressure = 1 atm External pressure = 1 atm External pressure = 1 atm Internal pressure = 1 atm Internal pressure = 2 atm Internal pressure = 3 atm Temperature = 200 K



Volume = 1 unit Temperature = 400 K



Volume = 1 unit Temperature = 600 K.

"Note: Volume in this model is recorded in anity rather than liters because 4 molecules of gas at the conditions given would occupy a very small space (-1 × 10⁻²¹ µL). The particles shown here are much larger compared to the space between them than actual gas particles.

Gas Variables Answers

Kirsten Franklin, Paul Muir, Terry Scott, Paul Yates

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General, Organic, and Biochemistry Study Guide Marcia L. Gillette, Ira Blei, Wendy Gloffke, George Odian, 2006-01-27 This study guide provides reader friendly reinforcement of the concepts covered in the textbook Features include Chapter outlines Are you able to Worked text problems Fill ins Test yourself Concept maps Can also be used for Blei and Odian's Organic and Hypersonic and High Temperature Gas Dynamics John David Anderson, 1989 This book is a self contained text for those students and readers interested in learning hypersonic flow and high temperature gas dynamics It assumes no prior familiarity with either subject on the part of the reader If you have never studied hypersonic and or high temperature gas dynamics before and if you have never worked extensively in the area then this book is for you On the other hand if you have worked and or are working in these areas and you want a cohesive presentation of the fundamentals a development of important theory and techniques a discussion of the salient results with emphasis on the physical aspects and a presentation of modern thinking in these areas then this book is also for you In other words this book is designed for two roles 1 as an effective classroom text that can be used with ease by the instructor and understood with ease by the student and 2 as a viable professional working tool for engineers scientists and managers who have any contact in their jobs with hypersonic and or high temperature flow Final Exam Review: Arithmetic A. A. Frempong, Arithmetic covers Basic Definitions Terminology and Types of Numbers Writing Whole Numbers Using Numerals and Words Basic Operations and Properties Order of Operations and Evaluation of Arithmetic Expressions Rounding off Whole Numbers and Decimals Estimation Prime Numbers Divisibility Rules Prime Factorization Least Common Multiple LCM Operations on Fractions and Mixed Numbers Addition and Subtraction of Fractions Comparison of Fractions and Subtraction of Mixed Numbers Multiplication and Division of Fractions and Mixed Numbers Operations on Decimals Comparison of Decimals Complex Decimals Dividing Decimals Converting Fractions to Decimals Ratio and Proportion Proportion Problems Percent % and Calculations Involving Percent Averages Profit and Loss Areas and Perimeters Bar Line and Circle Pie Graphs Scientific Notation Measurements

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climate conditions and has the dual role of contributing to global warming through emissions of the greenhouse gas methane Climate change has been recognized as a major threat to the global environment Because of insufficient field data rice growing countries face a problem when trying to comply with the United Nations Framework Convention on Climate Change stipulations to compile a national inventory of emissions and to explore mitigation options Given the expected doubling in rice production in Asia the need to evaluate the interaction between climate change and rice production is critical to forming a sound basis for future directions of technology developments by policy makers agriculturists environmentalists rice producers and rice consumers The present book comprises two sections The first part documents a comprehensive overview of the results achieved from an interregional research effort to quantify methane emission from major rice ecosystems and to identify efficient mitigation options This research report broadens understanding of the contribution of rice cultivation to methane emissions and clarifies that emissions are relatively low except in specific rice ecosystems and that these high emissions could be ameliorated without sacrificing yield The second section shows results from other projects that investigated the role of rice cultivators in field and laboratory approaches The findings represent inputs for future modeling approaches in the role of rice cultivators The expanded database generated by other projects is reflected in modeling efforts

Proceedings of the Ohio Gas Light Association Ohio Gas Light Association,1904 Scandinavian Journal of Gastroenterology ,1987 Trends in Oil and Gas Exploration United States. Congress. Senate. Interior and Insular Affairs,1972 Trends in Oil and Gas Exploration United States. Congress. Senate. Committee on Interior and Insular Affairs,1972

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