

6. Test for Co^{2+} and Al^{3+}

Some 6M HCl is added to the colorless solution. No precipitate forms. Next, some 3% H_2O_2 and 6M NaOH are added until the solution is basic, and then it is heated. No precipitate forms. The solution is acidified with 6M HNO_3 and then made basic with 6M NH_3 . No precipitate forms.

Ions present: _____ Ions absent: _____ Ions Undetermined: _____

7. Test for Ag^+ , Fe^{3+} , Bi^{3+} , and Zn^{2+}

Some 6M HCl is added to the colorless solution. A white precipitate forms and is removed. The solution is then made more basic with 6M NaOH. No precipitate forms.

Ions present: _____ Ions absent: _____ Ions Undetermined: _____

Only one of each of the following pairs of reactants undergoes a reaction. Complete and balance the equation for the reaction which occurs.



Qualitative Analysis Of Cations Experiment 19 Answers

**James M. Postma, Julian L. Robert, J.
Leland Hollenberg**



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