

Lubricants Cross Reference Guide

OEM PART #	Lubricant Type	Viscosity
York A	Mineral Oil/Capella	300 SUS/68 ISO
York C	Mineral Oil	300 SUS/68 ISO
York D	Mineral Oil	150 SUS/32 ISO
York E	Mineral Oil	500 SUS/100 ISO
York F	Mineral Oil	150 SUS/32 ISO
York G	Polyolester	1500 SUS/1320 ISO
York H	Polyolester	300 SUS/68 ISO
York J	Polyolester	200 SUS/46 ISO
York K	Polyolester	150 SUS/32 ISO
York L	Polyolester	600 SUS/120 ISO
York O	Polyolester	150 SUS/32 ISO
York P	Polyolester	750 SUS/150 ISO
York S	Polyolester	300 SUS/68 ISO
Trane 15	Pale Mineral Oil	300 SUS/68 ISO
Trane 33	White Mineral Oil	300 SUS/68 ISO
Trane 31	Pale Oil w/ Silicone	300 SUS/68 ISO
Trane 32	Pale Mineral Oil	150 SUS/32 ISO
Trane 37	Polyolester	300 SUS/68 ISO
Trane 42	Pale Oil w/ Phosphate	150 SUS/32 ISO
Trane 43	Pale Mineral Oil	150 SUS/32 ISO
Trane 45	Pale Mineral Oil	300 SUS/68 ISO
Trane 48	Polyolester	300 SUS/68 ISO
Trane 78	Polyolester	150 SUS/32 ISO
Mitsubishi 160 SZ	Polyolester	150 SUS/32 ISO

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MINERAL OIL	Part #	Viscosity	Equivalent Part #	OEM Part #
NL 150	1501G	150 SUS/32 ISO	Calumet C3, Suniso 3G5	York D, F – Trane 32, 43
NL 300	3001G	300 SUS/68 ISO	Calumet CA, Suniso 4G5	York C – Trane 15, 31, 45
NL 500	5001G	500 SUS/100 ISO	Calumet C5, Suniso 5G5	York E
NL WF 32	WF32	150 SUS/32 ISO	Texasco WF 32	
NL WF 68	1TD	300 SUS/68 ISO	Texasco WF 68	York A

ALKYLBENZENE	Part #	Viscosity	Equivalent Part #	OEM Part #
NL AKB 150	150AKB1G	150 SUS/32 ISO	Zenel 150, Solhex 150	
NL AKB 200R	200AKB1G	200 SUS/46 ISO	Zenel 200TE	
NL AKB 300	300AKB1G	300 SUS/68 ISO	Zenel 300, Solhex 300	

POLYOLESTER	Part #	Viscosity	Equivalent Part #	OEM Part #
NL PE 32	PE321G	150 SUS/32 ISO	Ultra 32-3MAE, Castrol SW32	York K, O – Trane 78
NL PE 68	PE681G	300 SUS/68 ISO	Emulacate 68RL, Castrol SW68	York H – Trane 37, 48

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Rachel S Tattersall



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methods in choosing installing maintaining troubleshooting servicing and repairing today s AC and refrigeration equipment
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web hexane is a nonpolar liquid with a dipole moment of zero and therefore does not significantly interact with the ions of the nacl crystals 11 a fe no 3 3 is a strong electrolyte thus it should completely dissociate into fe 3 and no 3 ions therefore z best represents the solution b fe no 3 3 s fe 3 a q 3 no 3 a q

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web here the value of concentration of water is taken as 0 36 instead of 0 16 divide the product of the concentration of water carbon monoxide and the equilibrium constant by the concentration of hydrogen instead of multiplying it correct d the value of K_{eq} is taken as 4 4 instead of 0 44