

Question: What happens to particles when heated?

TEMPERATURE AND HEAT

Temperature – a measure of the average value of the kinetic energy of the molecules in random motion. (SI Unit for temperature is Kelvin (K)).

Thermal expansion – almost all substances expand when they are heated and contract when they are cooled ----- exception water

Thermal energy – sum of the kinetic and potential energy of all the particles in an object; thermal energy of an object increases as temperature increases

Temperature Conversion Equations

$^{\circ}\text{F} \rightarrow ^{\circ}\text{C}$

$$^{\circ}\text{C} = (5/9)(^{\circ}\text{F} - 32)$$

$^{\circ}\text{C} \rightarrow ^{\circ}\text{F}$

$$^{\circ}\text{F} = (9/5)(^{\circ}\text{C}) + 32$$

Heat – thermal energy that flows from something at a higher temperature to something at a lower temp.

Specific heat – amt of heat needed to raise the temp of 1 kg of some material by 1°C

Specific Heat of Common Materials

Substance	Spec. Heat (J/(kg $^{\circ}\text{C}$))
Water	4,184
Wood	1,760
Carbon	710
Glass	664
Iron	450

Thermal Energy Equation

(Q) Change in thermal energy (J) =
 Mass (kg) x Δ temp ($^{\circ}\text{C}$) x
 (c) Specific heat (J/kg $^{\circ}\text{C}$)

$$Q = m(T_f - T_i) C$$

Assessment Chapter Test Thermal Energy Physical Science

Lei Shi



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