

3. A form is represented by a compound $\mathbf{A}$ 

Answer: 23

Pharmazie, 66, 2011, 10, 789–792

Substitution: $\frac{1}{2} \ln \left(\frac{1}{2} - \frac{1}{2} \right) = \frac{1}{2} \ln \left(\frac{1}{2} - 1 \right)$

mass of electron =

ISOTOPE : Atoms of the same element with the same proton number but a different neutron number

- Have the same properties due to the same number of electrons on the outer shell

Radioisotope:

- testing for leaks in pipes
- treating cancer

IONIC BONDING

Bonding: Oppositely charged ions are strongly attracted by electrostatic force
+ involves electron transfer

metal + non metal

- metals give electrons to non-metals

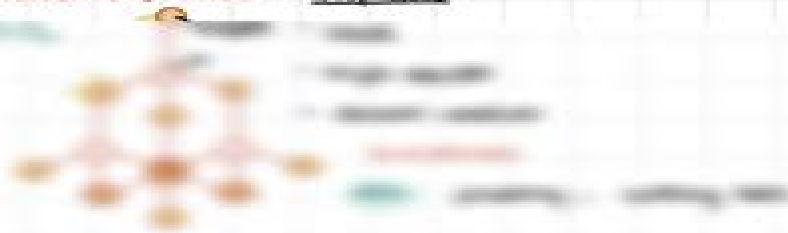
properties: strong bonds due to attractive forces

- conduct electricity
- charged particles
- High mp/bp
- Soluble in water

Small form of 4-cy 1,5-dicarbonyl, 4,5

SILICON (IV) OXIDE

Proposed Method



For example, when a company is reporting, stating that it has adopted an explicit policy to "do no harm" may be seen as a positive step. On the other hand,

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