

20. You have two solutions, A and B. The pH of solution A is 6 and pH of solution B is 8. Which solution has more H^+ ion concentration ?
- (1) A
 - (2) B
 - (3) Both have equal concentration
 - (4) Cannot be determined.
21. An aqueous solution turns red litmus solution blue. Excess addition of which of the following solutions would reverse the change ?
- (1) Baking powder
 - (2) Lime
 - (3) Ammonium hydroxide solution
 - (4) Hydrochloric acid
22. When a mixture of iron oxide and aluminium powder, is heated, intense heat is generated. This fact is made use of in
- (1) Soldering
 - (2) Welding of steel
 - (3) Making iron
 - (4) Cutting of metals
23. Which of the following statements is/are correct about iron?
- (A) Stainless steel is an alloy of nickel, chromium and iron.

25. How are the following related to each other ?
- (A) 3-Methylbutan-2-ol
 - (B) 3,3-Dimethylbutan-2-ol
 - (C) 2,3,3-Trimethylbutan-2-ol
- (1) They are allotropes.
 - (2) They are isomers.
 - (3) They are homologues.
 - (4) They are isobars.
26. An element X has electronic configuration 2, 8, 3. What is true about this element ?
- (A) Its atomic radius will be more than that of an element Y having electronic configuration 2, 8, 7.
 - (B) The oxide formed by X will react with aqueous solution of hydroxide of an element with electronic configuration 2, 8, 1.
 - (C) It lies in the same group as Boron (metalloid) and gallium.
- (1) A & B
 - (2) B & C
 - (3) Only B
 - (4) A, B & C
27. Which of the following properties generally increases on moving from top to bottom in a group ?
- (1) Ionization energy
 - (2) Non-metallic character

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