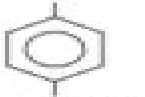


Unit 1 Organic Chemistry

ARE YOU READY?

(Page 4)

Understanding Concepts

- 3,3-dimethylpentane
 - 1-methyl-4-propylcyclohexane
- $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
 - $$\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_2\text{CCH}_2\text{CH}_2\text{CH}_2\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$$
 - 
 - $$\begin{array}{c} \text{CH}_2\text{CH}_3 \\ | \\ \text{CH}_3 \\ | \\ \text{CH}_2\text{CHCH}_2\text{CH}_2\text{CH}=\text{CH}_2 \end{array}$$
 - $\text{CH}_3\text{CH}_2\text{C}\equiv\text{CH}$
 - $\text{C}_7\text{H}_{16} + 11\text{O}_2 \rightarrow 7\text{CO}_2 + 8\text{H}_2\text{O}$
 - Cyclohexane (a) and 2,3-dimethyl-2-butene (d) are structural isomers; cyclohexene (b) and hexyne (c) are structural isomers.
 - Pentane would have a higher boiling point because it has a long carbon chain allowing more van der Waals attractions between molecules. The 2,2-dimethylpropane has a more spherical shape, with fewer intermolecular attractions. More energy is required to separate molecules with stronger intermolecular attractions, and thus, these compounds have higher boiling points.
 - 2,2-dihydroxyethane, (a), will have a greater solubility in water than does methylbenzene, (b).
 - H, (C, S), (N, Cl), O
 - intramolecular: single covalent bonds; intermolecular: van der Waals
 - intramolecular: single covalent bonds; intermolecular: van der Waals forces and hydrogen bonds
 - intramolecular: single covalent bonds; intermolecular: van der Waals forces and hydrogen bonds
 - Liquid 1 is 2-methylbutane; liquid 2 is 2-methyl-2-butene; liquid 3 is pentane.
 - The safety precautions needed are as follows: wear eye protection and a lab apron; avoid direct contact with skin (if in contact with skin, wash with copious amounts of water); when diluting concentrated acids, add acid to water; handle concentrated acids in a fume hood; store in glass or plastic container in secure location.
 - The safety precautions needed are as follows: wear eye protection and a lab apron; do not use with open flames; store in secure location.

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JS Bruner



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Exploring Ethnographic and Non-Ethnographic Approaches of Suicide and Self-Harm Elena Xen, 2019-07-22 This volume was first published by Inter Disciplinary Press in 2015 Ethnographic and non ethnographic approaches to suicide and self harming are explored in this volume With contributions from authors withr research teaching and practical experience in the field of suicide usually seen as a solution a response to the collapse of hopes plans and expectations and a complex death phenomenon surrounded by fear and taboo this volume attempts to add to little empirical research on suicide and self harm as the fastest growing behavioural problem amongst teenagers and adolescents

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Comprehensive Desk Reference of Polymer Characterization and Analysis Robert F. Brady, 2003 This book is a practical manual for those who analyze polymers Self contained chapters describe when a technique should be selected explain its basic principles describe how instruments are constructed and operated and teach how the data obtained relate to molecular structure and physical properties Many clear illustrations are included Implicit memory refers to a change in task performance due to an earlier experience that is not consciously remembered This book is not a research manual but rather a guide to performing and understanding polymer characterization and an introduction to the specialized literature of the analytical chemistry of polymers The techniques covered are directly relevant to the characterization of synthetic polymers such as adhesives sealants polymers composites coatings elastomers rubber and other nonmetallic materials Many techniques are also quite useful for natural and biological polymers

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