

Collected Lab Data

Mass of magnesium (g)	.211
Moles of magnesium (mol)	2.11
Temperature of water (°C)	25.0
Temperature of water (K)	298.15
Vapor pressure of water (mmHg)	23.8
Barometric pressure (mmHg)	755
Observations	
Bubbly upward force is the reaction.	
Volume of hydrogen gas collected (mL)	13.5
Volume of hydrogen gas collected (L)	0.0135
Pressure of hydrogen gas (mmHg)	.0135
Pressure of hydrogen gas (atm)	.990
Moles of hydrogen gas produced (mol)	102.4
Ideal gas constant (L·atm·mol ⁻¹ ·K ⁻¹)	401.5265

Indicates incorrect data

Determination Of The Ideal Gas Law Constant Lab

Ian Pickup



Determination Of The Ideal Gas Law Constant Lab:

Illustrated Guide to Home Chemistry Experiments Robert Bruce Thompson, 2012-02-17 For students DIY hobbyists and science buffs who can no longer get real chemistry sets this one of a kind guide explains how to set up and use a home chemistry lab with step by step instructions for conducting experiments in basic chemistry not just to make pretty colors and stinky smells but to learn how to do real lab work Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater bone and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints blood drugs and poisons and much more From the 1930s through the 1970s chemistry sets were among the most popular Christmas gifts selling in the millions But two decades ago real chemistry sets began to disappear as manufacturers and retailers became concerned about liability em The Illustrated Guide to Home Chemistry Experiments steps up to the plate with lessons on how to equip your home chemistry lab master laboratory skills and work safely in your lab The bulk of this book consists of 17 hands on chapters that include multiple laboratory sessions on the following topics Separating Mixtures Solubility and Solutions Colligative Properties of Solutions Introduction to Chemical Reactions Stoichiometry Reduction Oxidation Redox Reactions Acid Base Chemistry Chemical Kinetics Chemical Equilibrium and Le Chatelier s Principle Gas Chemistry Thermochemistry and Calorimetry Electrochemistry Photochemistry Colloids and Suspensions Qualitative Analysis Quantitative Analysis Synthesis of Useful Compounds Forensic Chemistry With plenty of full color illustrations and photos Illustrated Guide to Home Chemistry Experiments offers introductory level sessions suitable for a middle school or first year high school chemistry laboratory course and more advanced sessions suitable for students who intend to take the College Board Advanced Placement AP Chemistry exam A student who completes all of the laboratories in this book will have done the equivalent of two full years of high school chemistry lab work or a first year college general chemistry laboratory course This hands on introduction to real chemistry using real equipment real chemicals and real quantitative experiments is ideal for the many thousands of young people and adults who want to experience the magic of chemistry *Laboratory Experiments for Brown and LeMay, Chemistry, the Central Science* John Henry Nelson, Kenneth C. Kemp, 1977 **Chemistry in the Laboratory** James M. Postma, Julian L. Roberts, J. Leland Hollenberg, 2009-08-07 For nearly 40 years Chemistry in the Laboratory has been meeting the needs of teachers and students This new edition builds on that legacy while addressing cutting edge trends in the chemistry laboratory including forensic chemistry and environmental and green chemistry As always the new edition of Chemistry in the Laboratory offers precise easy to follow instructions helpful illustrations and an emphasis throughout on laboratory safety Again throughout a Consider This feature encourages users to expand the principles of the experiment into interesting applications open ended experiments or unexplored corners Most experiments in the manual can be completed in one lab session but some can be linked or extended for a multi lab

project **Exploring General Chemistry in the Laboratory** Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science **Bowker/Bantam ... Complete Sourcebook of Personal Computing** ,1984 **Instructors Manual to Lab Manual** Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 Introduction to Chemical Processes: Principles, Analysis, Synthesis Regina M. Murphy, 2007 Introduction to Chemical Processes Principles Analysis Synthesis enhances student understanding of the connection between the chemistry and the process Users will find strong coverage of chemistry gain a solid understanding of what chemical processes do convert raw materials into useful products using energy and other resources and learn about the ways in which chemical engineers make decisions and balance constraints to come up with new processes and products The author presents material and energy balances as tools to achieve a real goal workable economical and safe chemical processes and products Loaded with intriguing pedagogy this text is essential to a students first course in Chemical Engineering Additional resources intended to guide users are also available as package options such as ChemSkill Builder

The Software Directory for the APPLE Computer ,1981 *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *A Short Guide to Writing about Chemistry* Herbert Beall, John Trimbur, 1996 Advanced advice for students who want to read write and learn about chemistry in preparation for a career in that field Holt Physics Raymond A. Serway, 2002 **The Software Encyclopedia 2001** ,2001 **Fundamentals of Chemical Engineering** Charles Manson Thatcher, 1962 *Science Abstracts* ,1911 **An Algorithm for Determining Soil and Water Helium Concentrations from Headspace Analyses** Kim R. Green, 1984 Selected Water Resources Abstracts ,1986-07

Laboratory Experiments in Environmental Chemistry D. Neal Boehnke, R. Del Delumyea, 2000 This lab manual provides an interdisciplinary collection of 23 extensively tested environmental chemistry experiments with extensive introductory background material for each experiment It covers a broad range of methods and provides detailed instructions on calculation of results Experiments involve for example inorganic and organic profile of sediment and soil cores the pH of environmental waters and buffer capacity alkalinity of streams and lakes trace levels of ions in natural waters conductivity of natural waters chloride ion in natural waters colorimetry and absorption spectra metals in natural waters and in sediments atomic absorption spectrometry the chemical oxygen demand of natural waters and wastewaters the fluorimetric determination of polycyclic aromatic hydrocarbons environmental hydrocarbons air sampling particulates in urban air carbon

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