

KEY

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TEAM
①

Chemistry Lab Experiment

Determination of the Molar Volume of a Gas

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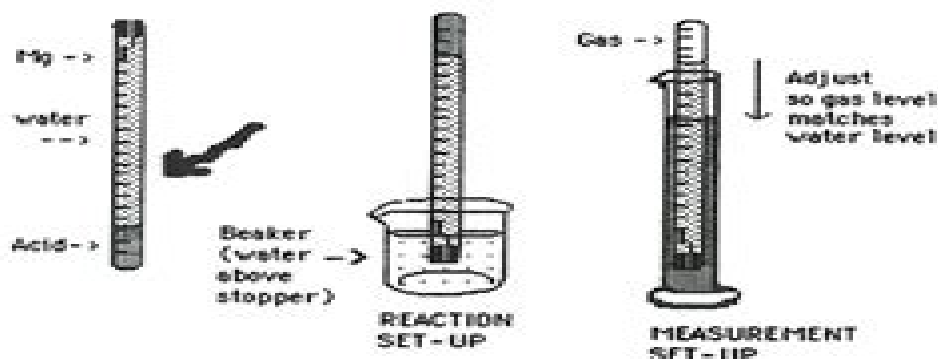
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Purpose:

- To experimentally determine the molar volume of hydrogen gas and to compare it with the $\frac{22.4\text{ L}}{\text{mole}}$ theoretical STP value.
- To relate gas law concepts learned in MA Frameworks Topic 6 (e.g., the Combined Gas Law and Dalton's Law of Partial Pressures) to previously learned concepts involving balancing chemical equations and stoichiometric relationships.

Introduction:



The **water displacement method** can be used to capture the hydrogen gas liberated when a strip of magnesium metal reacts with a $3.0\text{-}M$ solution of hydrochloric acid. The total gas volume of H_2O vapor and H_2 adjusted to the room's barometric pressure can be determined from the **eudiometer** (graduated gas-measuring tube) by partially immersing the tube into a large container of water until the liquid level inside the tube matches the water level in the large container.

According to **Dalton's Law of Partial Pressures**, the total gas pressure in the eudiometer is the sum of the pressure of the hydrogen gas collected plus the **water vapor pressure**. Symbolically, $P_{\text{total}} = P_{\text{H}_2} + P_{\text{H}_2\text{O}}$. Subtracting the water vapor pressure from the total pressure gives the pressure of the dry hydrogen gas.

Determining Molar Volume Gas Post Lab Answers

**United States. National Aeronautics
and Space Administration**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the NASA text box.

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Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

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with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations **Scientific and Technical Aerospace Reports** ,1992 *Nuclear Science Abstracts* ,1974 **Energy Research Abstracts** ,1992-10 *Technical Publications Announcements with Indexes* United States. National Aeronautics and Space Administration,1962 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration,1977 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration. Technical Information Center,1977 Code of Federal Regulations ,1992 **The Code of Federal Regulations of the United States of America** ,1990 The Code of Federal Regulations is the codification of the general and permanent rules published in the Federal Register by the executive departments and agencies of the Federal Government Radioactive Waste Processing and Disposal U.S. Nuclear Regulatory Commission,1980 **Technical Abstract Bulletin** , Microorganisms in Polar Regions: Understanding Their Survival Strategies for a Sustainable Future Prashant Kumar Singh,Shiv Mohan Singh,Trista J. Vick-Majors,2024-07-02 Scientific research on cold adapted microorganisms specifically polar microbes is of great interest since Arctic and Antarctic regions harbor diverse and active populations of microorganisms However these microorganisms are subject to impacts of environmental perturbations For example climate change will modulate the distribution and activity of many cyanobacteria and algal species in polar environments that contribute significantly to global carbon fixation and oxygen production Moreover many microorganisms that have remained frozen for thousands of years can revive their metabolic activity and re join the modern microbial community For survival in freezing environments polar microorganisms have established specific regulatory mechanisms which are now being challenged by new rapidly changing environmental conditions Remarkable progress has been made to uncover microbial adaptation to anthropogenic activities such as high irradiance nutritional deprivation UV B radiation heat cold desiccation and heavy metals Studies have enabled a basic understanding of gene regulatory pathways for morphological physiological metabolic and genetic adaptations to various environmental stresses To further elucidate physiological adaptation mechanisms and molecular diversity in polar regions this Research Topic is focused on microbes in polar regions looking at their biodiversity ecological adaptations the impact of climate change on their lifestyles and biotechnological application of microbes for a sustainable future Federal Register ,1979-12 *Government Reports Announcements & Index* ,1993-06

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